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THIOCARBONYL COMPLEXES OF

RUTHENIUM AND OSMIUM

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ABSTRACT

This thesis deals with thiocarbonyl complexes of ruthenium and osmium [L_nM -CS].

In chapter 1 a comprehensive review of all transition metal thiocarbonyl complexes is presented. Unlike carbon monoxide, carbon monosulphide is a highly reactive molecule which does not exist under normal laboratory conditions. However, transition metal complexes of this reactive intermediate are typically stable compounds of which there are now approximately two hundred reported examples. This small number has made a complete review possible. Slow spasmodic growth has accompanied transition metal thiocarbonyl chemistry since the prototype complex was discovered in 1966 and chapter 1 discusses the synthetic reasons for this, describes critically the current bonding models, and presents the reactions The very new field of the chemistry of thiocarbonyl complexes. of selenocarbonyl complexes $[L_n^{M-CSe}]$ is included and comparison with this and with the well-documented chemistry of carbonyl complexes allows for an analysis of periodic changes.

The review indicates that CS ligand reactions had received little attention at the outset of this project. Interligand reactions involving CS were unknown primarily because thiocarbonyl complexes containing ligands suitable for investigating reactions of this type had not been synthesised. In particular, transfer of hydrido or aryl ligand to CS had not been observed. Such

reactions are the subjects of chapters 3 and 6. The investigation of these transfer reactions necessitated the prior preparation of complexes containing hydrido or aryl ligands cis to the thiocarbonyl ligand. Suitable complexes were synthesised from the very versatile synthetic intermediate OsCl₂(CS)(PPh₃)₃ [Chapter 3], which was discovered after considerable effort had been devoted to finding high yield routes for introducing the CS ligand into the coordination sphere of ruthenium and osmium [Chapter 2].

Chapter 3 is concerned with hydride transfer to the CS ligand. The thioformyl ligand $[Os-C]_S^H$ results from the transfer of one hydrido ligand to CS and subsequent reactions, which are also probably hydride transfers, produce the thioformaldehyde

[OS I and methylthiolato [OS-SCH3] ligands. The possible relevance of these ligands as models for postulated intermediates in the Fischer-Tropsch synthesis is noted. The thioformyl and thioformaldehyde ligands were hitherto unknown and the formulations have been additionally verified by investigating the reactivity of these novel ligands and characterising the derivatives obtained. A synthetic route to secondary carbene complexes and to nitrogen and oxygen equivalents of the thioformyl ligand [OS-CNMe] and OS-CSONME

The review in chapter 1 also recounts how observation has shown that strong σ -donor and strong π -acceptor properties of the CS ligand exert a profound effect upon the reactivity of CS complexes. In particular, strong π -acceptor properties mark the CS ligand as an ideal candidate for stabilising very electron-rich metal centres

such as d⁸ ruthenium(O) or osmium(O). In chapter 4 high yield routes to zerovalent osmium thiocarbonyl complexes are detailed and the reactivity of these complexes is exploited to afford the first high yield syntheses of dithiocarbonyl complexes.

The thiocarbonyl ligand is also viewed, from the collected information in chapter 1, as being more responsive to the coordinative environment than the better-known carbonyl ligand.

Thus, while CS can remove more electron-density from basic metal centres than CO, CS can also donate more electron-density than CO when coordinated to very electron-poor metal centres.

Consequently, it is not surprising that reactions which have been regarded in a classical sense as electrophilic or nucleophilic attack at CO are also found for CS. In chapter 5 several novel reactions which can be described as nucleophilic attack at the CS carbon atom are reported for both neutral and cationic complexes.

One such reaction is an interligand reaction of cis thiocarboxamido and thiocarbonyl ligands which affords a four-membered osmium metallocycle.

Chapter 6 describes how coordinatively unsaturated arylthiocarbonyl-containing complexes react with potentially bidentate anionic ligands to produce coordinatively saturated complexes.

Coordinative saturation also occurs with carbon monoxide or isocyanide and these latter compounds undergo a rearrangement reaction which involves the thiocarbonyl ligand. Available evidence, which includes the results of an incomplete X-ray crystal structure determination, suggests that a bidentate thioacyl ligand [Os \ \frac{C^R}{S} \] is produced by transfer of the aryl ligand to CS with ring-closure. Some further reactions of this novel ligand are also discussed.

In conclusion, the work reported here extends this research group's interest in the chemistry of low valent transition metal compounds and the ligands which stabilise such compounds to thiocarbonyl complexes of ruthenium and osmium. The present status of the chemistry of transition metal thiocarbonyl complexes has been examined and the work presented in the experimental chapters contributes to this small area of chemistry by examining, for the first time, interligand reactions which involve CS, by providing synthetic routes to CS complexes of osmium(0), and by beginning to examine the synthetic potential of the novel compounds and ligands obtained.

TABLE OF CONTENTS

	Page
ABSTRACT	i
List of Tables	viii
List of Important Figures	ix
List of Abbreviations	х
CHAPTER 1: INTRODUCTION	
THIOCARBONYL AND SELENOCARBONYL COMPLEXES OF	
TRANSITION METALS	_ 1
(I) SYNTHESIS:	1
 (i) Coordination of Molecular Carbon Monosulphide (ii) Interaction Between Carbon Disulphide and Carbon Diselenide and a Suitable Metal Substrate, Usually in the Presence of 	e; 3
Triphenylphosphine; (iii) Reductive Fragmentation of Thiophosgene; (iv) Modification of Thioester and Selenoester	6 15
Ligands; (v) Reductive Elimination of Methyl Thiol.	17 20
(II) STRUCTURE AND BONDING:	21
(i) X-Ray Crystallography;(ii) Infrared and Raman Spectroscopy;(iii) Theoretical Studies and Photoelectron	22 26
Spectroscopy; (iv) 13C N.M.R. Spectroscopy; (v) Mass Spectrometry;	3 3
(vi) Mössbauer Spectroscopy,	36 T
(III) CHEMICAL REACTIVITY	40
A. Reactions at the Metal Centre.	40
(i) Substitution Reactions;(ii) Oxidative Addition and Redox Reactions.	40 53

1	3,	Ligand	Reactions.	57
		(i) (ii) (iii) (iv)	Nucleophilic Attack at the CS Carbon Atom; Electrophilic Attack at the CS Sulphur Atom; Migratory Insertion Reactions; Intramolecular Isomerisation Reactions	57 62 64
		(v)	Involving Bridged CS Ligands; Miscellaneous Reactions of Other Than The CS Ligand.	67 69
(CONC	LUSION		69
CHAPT	TER	2		
	OUC		L COMPLEXES FROM CARBON DISULPHIDE	70
2.1		Thioca	reparative Routes To Ruthenium and Osmium arbonyl Complexes;	70
2.2		Osmium	To Find New Routes to Ruthenium and CS complexes;	73
2.3	A A	Novel	ment of Synthetic Routes to CS Complexes; Synthetic Route to Zerovalent Ruthenium	75
2.5		and Os	smium Complexes; ated Tetrafluoroborate and Perchlorate Complexes.	83
GEN	ERA	L EXPER	RIMENTAL	92
EXP	ERI	MENTAL		95
CHAPT	ER	3		
RED	UCI	CION O	F THE THIOCARBONYL LIGAND	116
3.1 3.2			(PPh ₃) ₃ ; Thiocarbonyl Complexes;	120 122
3.3 3.4		eaction (i)	Transfer Reactions; s of Thioformyl Complexes. Carbene Formation and Derived Products; Further Reduction of the Thioformyl Ligand.	128 132 137
EXP	ERII	MENTAL		143
CHAPT	ER	4		
ZER	.OVA	LENT	THIOCARBONYL COMPLEXES OF OSMIUM	168
4.1 4.2	Sy	ynthesi ynthesi	s and Reactions of Zerovalent Complexes; s of Dithiocarbonyl Complexes.	169 174
EXP	ERIN	MENTAL		179

CHAPTER 5	
NUCLEOPHILIC ATTACK AT THE THIOCARBONYL LIGAND	194
EXPERIMENTAL	200
CHAPTER 6	
ARYL-THIOCARBONYL COMPLEXES AND REARRANGEMENT	
REACTIONS	211
6.1 Production of Aryl-Thiocarbonyl Complexes;6.2 Rearrangement Reactions of Aryl-Thiocarbonyl Complexes.	212 215
EXPERIMENTAL	222
REFERENCES	237
ACKNOWLEDGEMENTS	252

LIST OF TABLES

		Pag
CHAPT	ER 1	
1.	Representations of CS and CSe Complex Structures	25
2.	I.R. Data for CSe Complexes and CS Analogues.	312
3.	13 C N.M.R. Data for Thiocarbonyl Complexes	38
CHAPT	ER 2	
2.1	I.R. Data for Thiocarbonyl Complexes	110
2.2	I.R. Data for Thioester and Related Complexes	111
2.3	I.R. Data for Other Complexes	112
2.4	¹ H N.M.R. Data	113
CHAPTI	ER 3	
3.1	I.R. Data for Hydrido-Thiocarbonyl Complexes	160
3.2	1 H N.M.R. Data for Hydrido-Thiocarbonyl Complexes	161
3.3	I.R. Data for Thioformyl and Derived Complexes	163
3.4	H N.M.R. Data for Thioformyl and Derived Complexes	165
3.5	I.R. and ¹ H N.M.R. Data for Related Adducts of	
	Os(CO) ₂ (PPh ₃) ₂	141
CHAPTE	CR 4	
4.1	I.R. Data for Thiocarbonyl Complexes	191
4.2	1 H N.M.R. Data for Thiocarbonyl Complexes	192
CHAPTE	R 5	
5.1	I.R. Data for New Complexes	207
	1 H N.M.R. Data for New Complexes	
CHAPTE		208
CHAPTE		
6.1	I.R. Data for New Compounds	232
6.2	H N.M.R. Data for New Complexes	234

LIST OF IMPORTANT FIGURES

	Pag
CHAPTER 2	
2.2 Structure of $[Ru(\eta^2-CS_2Me)(CO)_2(PPh_3)_2]^+$.	80
2.3 (a) The Inner Coordination Sphere of OsH(η^1 -CS ₂ Me)(CO) ₂ (PPh ₃) ₂	
with some distances presented.	86
(b) A stereoview of OsH(η^1 -CS ₂ Me)(CO) ₂ (PPh ₃) ₂ .	86
2.5 Structure of [RuH(H ₂ O)(CO) ₂ (PPh ₃) ₂]BF ₄ .	91
CHAPTER 5	
5.1 Structure of Os (η¹-CS ₂ CNMeR) H (CO) (PPh ₃) ₂ .	198
CHAPTER 6	
6.1 Preliminary Results of X-Ray Crystal Structure	
Analysis of OsBr(η^2 -CSR)(CO)(PPh ₃) ₂ .	217

LIST OF ABBREVIATIONS

ру

pyridine

phen

o-phenanthroline

bypy

2,2'-bipyridine

dmf

N, N-dimethylformamide

diphos

1,2-bis(diphenylphosphino)ethane

triphos

bis(2-diphenyphosphinoethyl)phenylphosphine

Ср

pentahapto-cyclopentadienyl

diars

o-phenylenebis(dimethylarsine)

TCNE

tetracyanoethylene

THF

tetrahydrofuran

acac

acetylacetonato

ØH

benzene

COD

cyclo-octa-1,5-diene

triflate

trifluoromethanesulphonate

DMSO

dimethylsulphoxide (perdeutero- in ¹H N.M.R. tables)

Ме

methyl

ethyl

nPr

n-propyl

Ph

phenyl

Су

Et

cyclohexyl

n_{Bu} or Bu

n-butyl

 $t_{\rm Bu}$

t-butyl

CNR

p-tolylisocyanide (unless otherwise specified)

conc.

concentrated