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Gelation of oxidised cereal beta-glucan extracts

Introduction

Cereal β -glucan, which occurs in plant cell walls, is the major component of soluble dietary fiber in oat and barley. Soluble fibers with a high solution viscosity, which is positively correlated to molecular weight (MW), lower the serum glucose and cholesterol levels. The relevant health claims on β -glucan as a functional food have been allowed by regulatory authorities, including the EFSA and the US FDA. In most processed-food systems, the native hydrolytic enzymes are inactivated, but oxidative degradation of this non-starch polysaccharide can, however, adversely affect the claimed health benefits. (Kivelä 2011). Although the physiological benefits are positively correlated with the viscosity of β -glucan, the relationship with its gelling capacity is not fully understood.

The scope of the Master's thesis included (1) the study of gelation characteristics of oxidised solutions of oat and barley β -glucan, and (2) correlating the results obtained by dynamic light scattering (DLS) microrheology with conventional rheology.

Methods

Native oat and barley β -glucans (OBG and BBG) were extracted, purified and oxidised, as per the procedure outlined in Fig. 1.

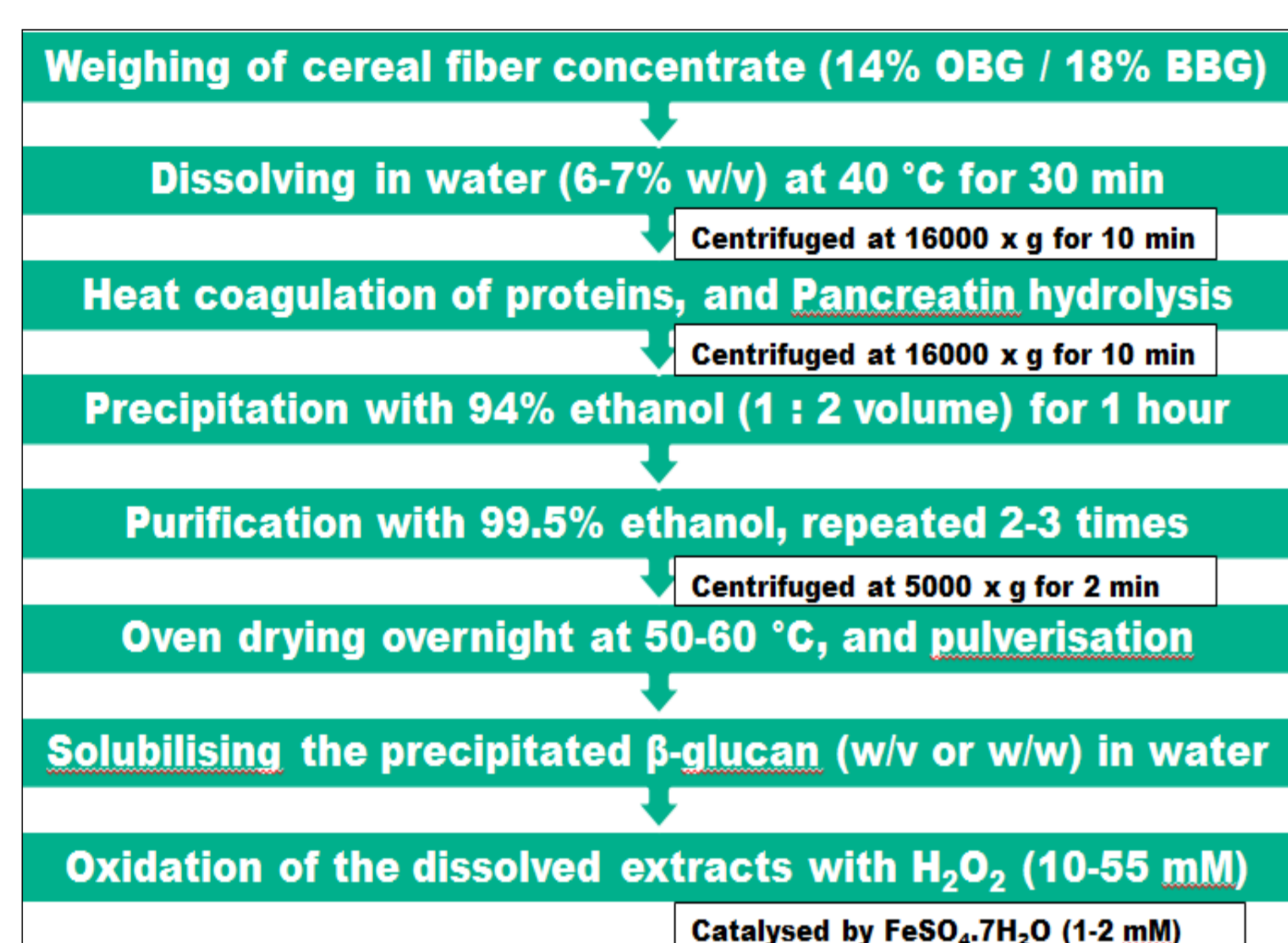


Fig. 1: Extraction and oxidation of β -glucan.

Rheological (RheoStress 600, Thermo Haake) and microrheological (Zetasizer Nano – ZS, Malvern) investigations were conducted on β -glucan solutions, 1-8 days after oxidising.

Results

Moduli curves of high-concentration BBG (4.0% w/w) illustrates the transformation in its

viscoelastic characteristics on oxidation (Fig. 2). Initially G'' was dominating on G' , and both the moduli increased with ω . Gelation was observed in the oxidised samples after a week, since G' became partially independent of frequency and predominated over G'' . Contrastingly in the non-oxidised samples, G' exceeded G'' initially, but there was an early cross-over and the viscous modulus prevailed at higher frequencies.

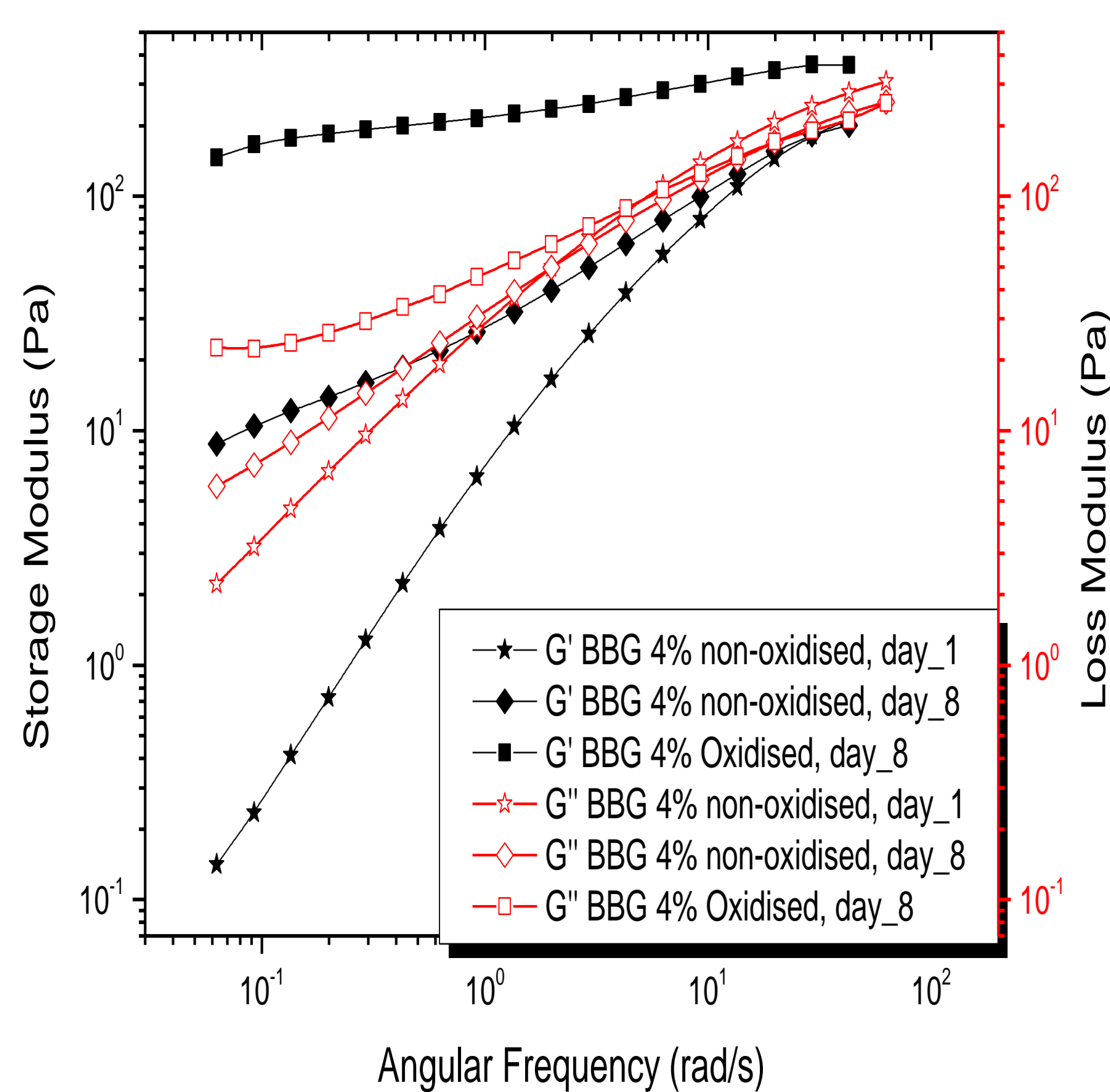


Fig. 2: Storage (G') and viscous (G'') modulus plotted against angular frequency (ω), for oxidised BBG and control samples.

Thixotropy was conspicuous in the gelling BBG solutions (Fig. 3). At the beginning, the viscosity curves with ascending shear overlap with those obtained with descending shear but after a week's storage, hysteresis between the up and the down curves was appreciable, particularly for the oxidised samples.

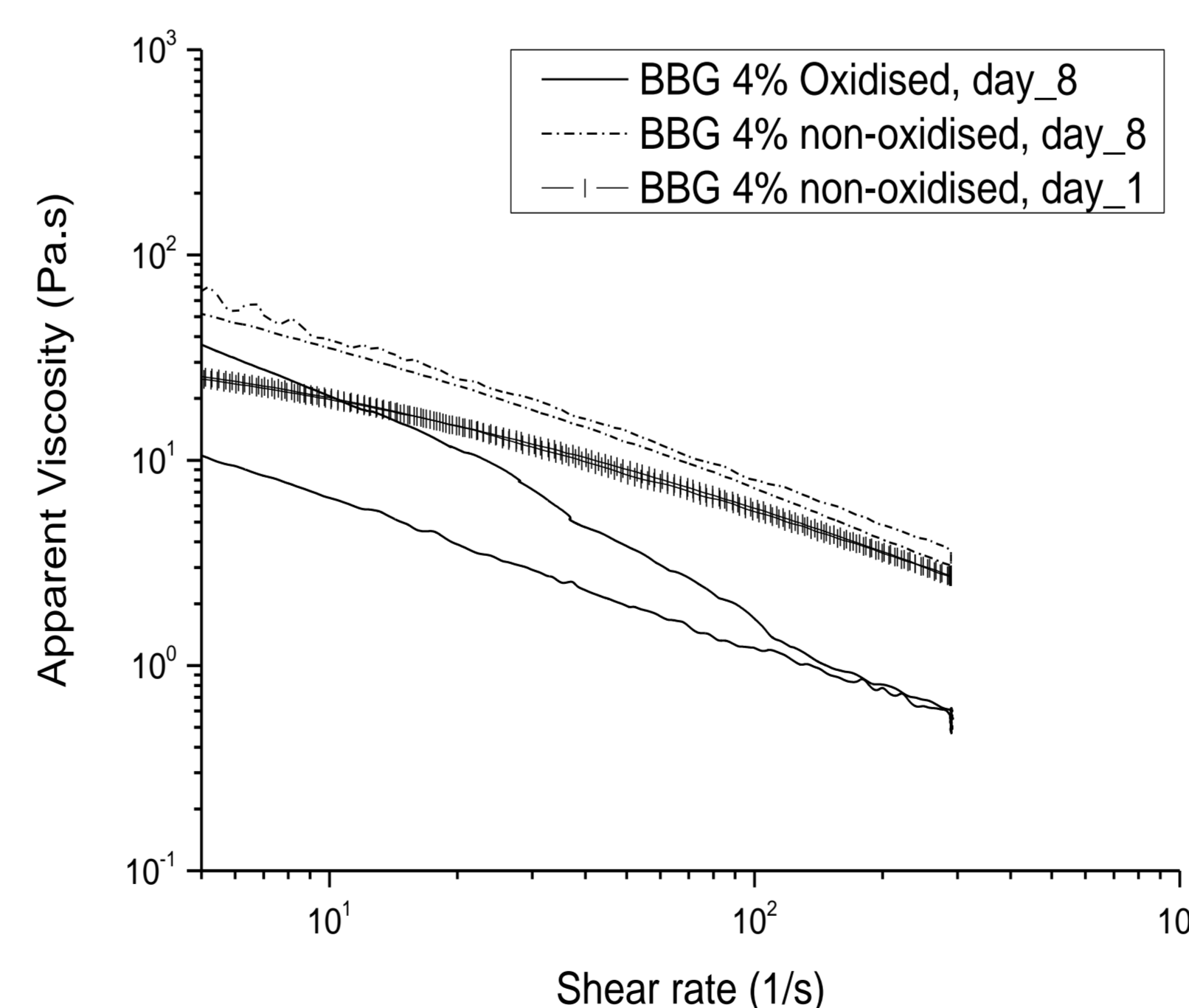


Fig. 3: Thixotropic loops for oxidised BBG and control samples.

In microrheology of dilute BBG solutions (0.2% w/v), η^* of the oxidised samples first decreased, but surged after a week with a shear-thinning slope (Fig. 4). Besides, G' gained in strength and became less dependent on ω , as also seen in the bulk rheology of high-conc. BBG (Fig. 2).

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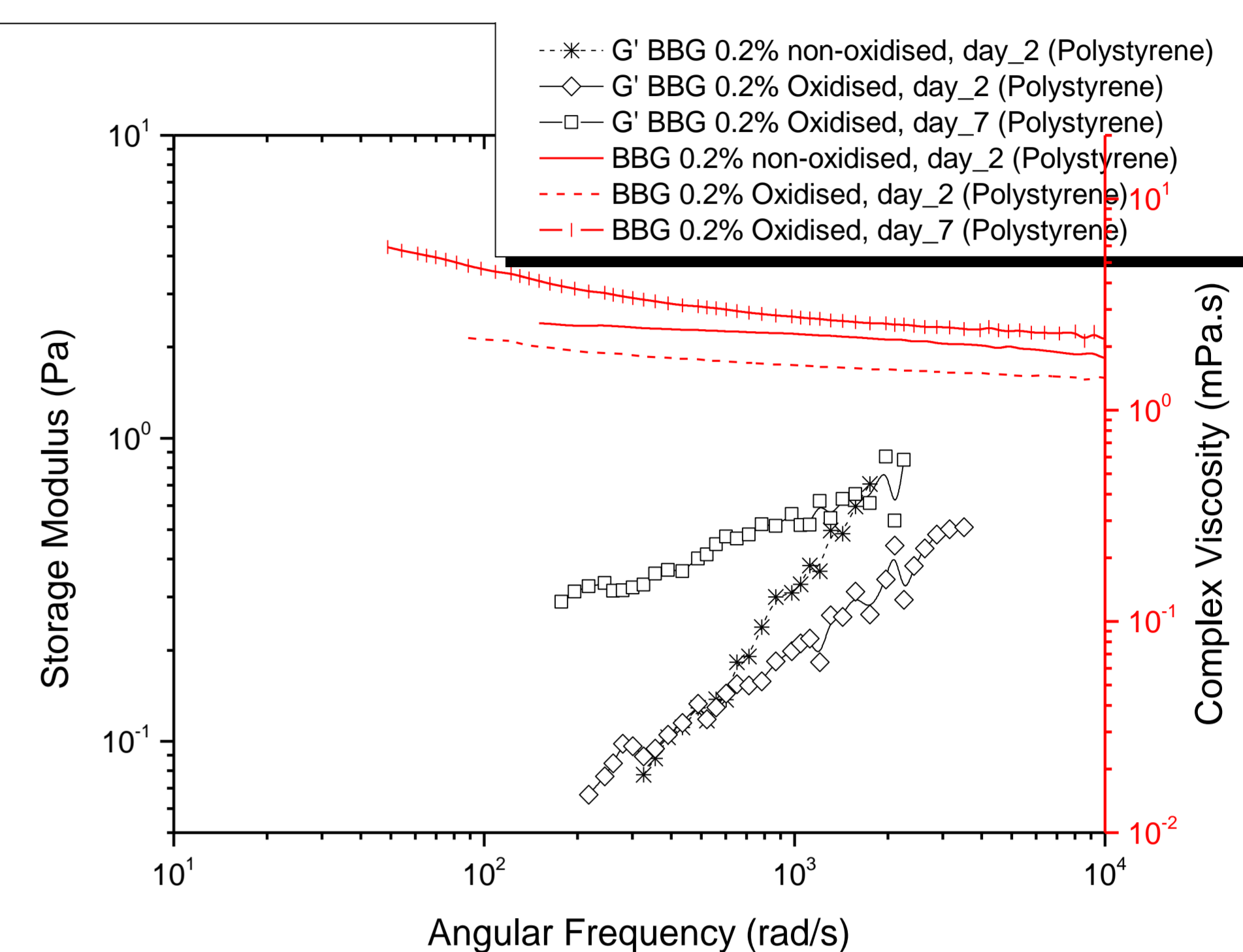


Fig. 4: Microrheology assessment with polystyrene probes in BBG.

Discussion

The low-MW BBG extract in the present study, which further depolymerised on oxidation, gelled more convincingly (Fig. 2) than the very high-MW OBG extract which somewhat forfeited its viscoelasticity due to oxidative degradation (results not shown for oat). This is also indicated in the microrheology data that are obtainable at a much higher frequency range (Fig. 4). In a study of the gelling properties of BBG extracts (8% w/v) that were acid-hydrolysed to different molecular weights, Vaikousi *et al*, 2004, also noted that the least amount of time was required by the lower MW polymers to acquire a gel structure. In addition to the effect of MW, the higher cellotriose / cellotetraose (DP3/DP4) ratio in BBG was also hypothesised to facilitate gelation. Furthermore, hysteresis due to rotational shear was more prominent in the gelling BBG (Fig. 3) than in OBG solutions.

Moschakis *et al*, 2012, observed an overlap in the values of G' , G'' and η^* for 2-5.0% (w/w) BBG extracts, obtained by rheology and direct imaging-based microrheology techniques. However, the different concentration regimes for bulk rheology and DLS microrheology in the present study make such an overlap unlikely.

Conclusion

Oxidised barley β -glucan showed the highest tendency for gel formation in oscillatory and rotational rheology, and also in microrheology experiments. This reaffirms that in cereal β -glucans, a low molecular weight and a high DP3/DP4 ratio is more conducive for gelation.

References

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Moschakis T, Lazaridou A, Biliaderis CG. 2012. *J Colloid Interface Sci* 375(1):50-59.
Vaikousi H, Biliaderis CG, Izydorczyk MS. 2004. *J Cereal Sci* 39(1):119-37.