



<http://researchspace.auckland.ac.nz>

### *ResearchSpace@Auckland*

#### **Copyright Statement**

The digital copy of this thesis is protected by the Copyright Act 1994 (New Zealand).

This thesis may be consulted by you, provided you comply with the provisions of the Act and the following conditions of use:

- Any use you make of these documents or images must be for research or private study purposes only, and you may not make them available to any other person.
- Authors control the copyright of their thesis. You will recognise the author's right to be identified as the author of this thesis, and due acknowledgement will be made to the author where appropriate.
- You will obtain the author's permission before publishing any material from their thesis.

To request permissions please use the Feedback form on our webpage.

<http://researchspace.auckland.ac.nz/feedback>

#### **General copyright and disclaimer**

In addition to the above conditions, authors give their consent for the digital copy of their work to be used subject to the conditions specified on the Library Thesis Consent Form.

**THERMAL EVOLUTION AND FLUID-ROCK INTERACTIONS  
IN THE ORAKEIKORAKO-TE KOPIA GEOTHERMAL SYSTEM,  
TAUPO VOLCANIC ZONE, NEW ZEALAND.**

**Gregory Bignall**

**February 1994**

**A thesis submitted in fulfilment of the requirements for the  
degree of Doctor of Philosophy**

**University of Auckland**

**1994**

## ABSTRACT

The active Orakeikorako-Te Kopia geothermal system was drilled in the mid-1960's, down to 1405m, as part of a programme to investigate its electrical generation capability. Four wells were completed at Orakeikorako (23km NNE of Taupo) and two at Te Kopia, 9.5km further northeast. The exploration drilling provided information on the present day hydrological and thermal regime which is as hot as 265°C (1137m drilled depth (-801m RL) in OK-2). Major flows into the wells occurred at depths down to 850m, although poor permeability and decline in mass output discouraged development. The waters discharged were of near neutral pH and had low salinities (highest Cl content from OK-2 ≈546mg/kg), low discharge enthalpies and indicated water temperatures ( $T_{SiO_2}$  and  $T_{NaKCa}$ ) of 210°C to 240°C.

A hydrologic model proposed here envisages a hot water reservoir in the OK-2 area (northeastern part of the Orakeikorako thermal area) with a lateral flow supplying water to the Red Hill (OK-4 area) in the southern part of the system and a concealed northeast flow which reaches the surface at Te Kopia.

The Orakeikorako thermal area occupies a surface area of about 1.8km<sup>2</sup>, mainly on the east bank of the Waikato River, where dilute chloride-bicarbonate water discharges along faults and fractures in association with an extensive silica sinter sheet, boiling springs and geysers. The occurrence of a mordenite-smectite assemblage at shallow depths, plus the oxygen and hydrogen isotopic composition of surface discharge waters, indicate that the ascending chloride fluids are diluted by near surface (heated?) groundwaters. The  $\delta D$  shift from local groundwater composition may be evidence for a magmatic component to the convecting hydrothermal system. Incursion of fluids from the relatively cool (<120°C) steam-heated carapace into deep levels of the system and its mixing with the alkali-chloride fluids, produces argillic alteration, sealing and will eventually result in the demise of the system. Old silica sinter on the west bank, at the foot of the Tutukau Rhyolite Dome, is covered by Oruanui Ash (22,700 years B.P.) and demonstrates that the hydrologic character of this part of the system has changed due to a combination of fault movement, changes in the height of the watertable and sealing.

The Te Kopia thermal area is located along 2.5km of the Paeroa Fault scarp, a major structural feature inferred to be controlling migration of deep hydrothermal fluids in the Orakeikorako-Te Kopia area. Surface activity at Te Kopia is characterised by acid alteration (including fumaroles, warm acid pools and steaming ground), although neutral pH alkali-chloride fluids discharged here within the last 3000 years and deposited silica sinter ( $C^{14}$  age on wood enveloped by sinter is 3026 +/- 43years B.P.). New thermal areas in the past year have begun to develop in the northwestern part of Te Kopia, whilst cold hydrothermally altered ground (hosting a mordenite + clay assemblage) records a decline in activity in the southern part of the Te Kopia thermal area.

The system is hosted by a generally SE dipping sequence of Pliocene to Quaternary ignimbrite, tuff and rhyolite lavas of the Taupo Volcanic Zone. Point counting, electron microprobe analyses of surviving primary phases (Fe, Mg, Al and Ti contents of hornblende and biotite), together with X-ray fluorescence and neutron activation analysis were used to distinguish three extensive ignimbrites encountered in the Orakeikorako-Te Kopia drillholes: (0.33Ma, sanidine-bearing) Paeroa, (0.35Ma) Te Kopia and (undated) Akatatarewa Ignimbrites, despite their having been hydrothermally altered. Ignimbrite recognition is made on the basis of a combination of immobile trace and rare earth element abundances and ratios: Ta and Yb (ppm),  $P_2O_5$ (%) and the ratios Zr/Yb, Zr/Y, Yb/Hf, La/Lu, Nb/Hf, Zr/Nb, Zr/Lu, Yb/Ta, Ta/Lu, La/Tb and Nb/Ta.

e.g.	<u>Paeroa Ignimbrite</u>	<u>Te Kopia Ignimbrite</u>	<u>Akatatarewa Ignimbrite</u>
Zr/Y	4.89 - 6.35 (mean 5.56)	9.67 - 11.92 (mean 10.78)	7.55 - 9.22 (mean 8.21)
Yb/Ta	2.84 - 3.38 (mean 3.11)	3.22 - 3.67 (mean 3.39)	3.84 - 5.11 (mean 4.32)
$P_2O_5$	0.013 - 0.040 (mean 0.026)	0.060 - 0.075 (mean 0.068)	0.070 - 0.105 (mean 0.087)

The Paeroa Ignimbrite is distinguished by its Eu anomaly  $((\text{Eu}/\text{Eu}^*)_{\text{cn}})$  is 0.48 to 0.54), whereas the Te Kopia and Akatarewa Ignimbrites are characterised by their flatter REE Spidergrams  $((\text{Eu}/\text{Eu}^*)_{\text{cn}})$  is  $\sim 1.0$  and  $\sim 0.8$  respectively). The correlation of the extensive ignimbrites was satisfactorily effected by a combination of their characteristic bulk rock and pumice chemistry, plus primary mineralogy, to enable the subsurface stratigraphy and structure of the Orakeikorako-Te Kopia geothermal system to be defined.

The alteration assemblage below 500m consists of quartz, albite+adularia, with variable abundance and distribution of chlorite, pyrite, calcite, wairakite, epidote, pyrrhotite, titanite, leucoxene, siderite and clinozoisite; illite is a late overprint. Rare almandine occurs in rhyolite in OK-1: 1312.5m; this is the first known occurrence of garnet in an active geothermal system of the TVZ. The surficial alteration assemblage of kaolin, cristobalite, alunite, hematite and jarosite reflects alteration by acid sulphate-steam heated waters.

The occurrence and textural relations exhibited by the hydrothermal mineral assemblage define the geochemical structure and thermal evolution of the Orakeikorako-Te Kopia system. The activities of components of minerals, determined from electron microprobe analyses and composition-activity relationships (OK-2 discharge:  $\log \alpha_{\text{K}^+}/\alpha_{\text{H}^+}=3.6$ ,  $\log \alpha_{\text{Na}^+}/\alpha_{\text{H}^+}=4.8$ ) support the petrologic observation that illite is now the stable potassium phase (overprinting adularia), although a state of equilibrium between the sheet silicate and the fluids is clearly not fully reached. The Na/K ratio of the altering fluid is controlled by the albite-adularia reaction (dissolution of albite and replacement by adularia, after andesine), whilst the  $\text{H}_2/\text{H}_2\text{S}$  ratio is buffered by the virtually complete replacement of pyrrhotite by pyrite. The fluids are now slightly undersaturated with respect to calcite, this is shown by etched surfaces on some calcite grains.

In the past the deep fluid boiled adiabatically from  $>300^\circ\text{C}$  to  $\sim 250^\circ\text{C}$  as it ascended, resulting in the deposition of adularia, quartz and bladed calcite. The system has cooled, resulting in lower subsurface temperatures (as recorded by fluid inclusion geothermometry) suppressing boiling, and migrated northwards as a consequence of self sealing. The thermal decline and retention of  $\text{CO}_2$  in the deep alkali-chloride fluid shifted the alteration assemblage from one of albite-adularia stability to illite stability.

The homogenisation ( $T_{\text{h}}$ ) temperatures of primary and secondary liquid-rich inclusions in 27 cores from different depths mostly match measured temperature profiles (e.g. OK-1 (shallow levels) and OK-2). Never-the-less, fluid inclusion data support mineral-inferred stability temperatures which indicate that parts of the Orakeikorako-Te Kopia system have cooled appreciably (e.g. OK-1, deep levels) and OK-4 (maximum  $T_{\text{bore}}=238^\circ\text{C}$ , maximum  $T_{\text{h}}=312^\circ\text{C}$ ; epidote abundant). In contrast, the northwestern margin (OK-6 area) has heated (OK-6:1113.4m;  $T_{\text{bore}}=261^\circ\text{C}$ ,  $T_{\text{h}}=210\text{-}221^\circ\text{C}$ ). Some inclusions in the Te Kopia drillholes have  $T_{\text{h}}$  values that exceed  $T_{\text{bore}}$  by as much as  $50^\circ\text{C}$ , and are deduced to have been uplifted by movement on the Paeroa Fault. Freezing data indicate that the trapped fluid was dilute ( $\sim 0.2$  to  $1.7$  wt% NaCl equivalent) since most  $T_{\text{m}}$  values range from  $-0.1$  to  $-0.5^\circ\text{C}$ .

The outflow portion of the Orakeikorako-Te Kopia system has evolved recently, both chemically and physically. Movement on the Paeroa Fault, that uplifted pyroclastic rocks hosting a quartz-adularia-illite assemblage, combined with a lowering of the watertable has resulted in an overprinting of the neutral pH hydrothermal mineral assemblage by a kaolinite-alunite type assemblage which derives from an acid sulphate fluid. Quartz crystals found 150m above the base of the Paeroa Fault scarp host dilute ( $\sim 1.5$ wt% NaCl equivalent) fluid inclusions with  $T_{\text{h}}$  values that range from  $180\text{-}206^\circ\text{C}$  (average  $196^\circ\text{C}$ ). Bladed quartz (after calcite) did not contain usable inclusions. It is deduced that the inclusions formed about 120-160m below the ground, which indicates uplift in the order of  $\sim 300$ m. Assuming a constant rate of uplift of 4m/ka (based on the offset of 330ka Paeroa Ignimbrite), the minimum duration of activity at Te Kopia is 75,000 years.

## CONTENTS

Abstract .....	i
Table of Contents .....	iii
List of Figures .....	xi
List of Tables .....	xiv
Acknowledgements .....	xv
<b>Chapter 1 :</b>	
<b>Active Hydrothermal Systems</b>	
<b>Nature of active systems in the Taupo Volcanic Zone</b>	<b>1</b>
1.1	Introduction .....
1.2	Volcanic and Structural Setting .....
1.3	Characteristics of Geothermal Systems of the TVZ .....
1.4	Fluid Composition and Sources .....
1.5	Geothermal Alteration Mineralogy (Distribution and factors controlling alteration) .....
1.5.1	Temperature .....
1.5.2	Pressure .....
1.5.3	Rock type .....
1.5.4	Permeability .....
1.5.5	Fluid composition .....
1.5.6	Duration .....
1.6	Style of Alteration Mineralogy .....
<b>Chapter 2 :</b>	
<b>Geology of the Orakeikorako-Te Kopia Area,</b>	
<b>Taupo Volcanic Zone</b>	<b>14</b>
2.1	Introduction - Location .....
2.2	History of Investigations at Orakeikorako-Te Kopia .....
2.2.1	Orakeikorako area .....
2.2.2	Te Kopia area .....
2.3	Regional Tectonic and Volcanic Setting .....
2.3.1	Structural features in the Orakeikorako-Te Kopia area .....
2.3.2	Maroa Volcanic Centre .....
2.3.3	Basement .....
2.4	Geology of the Orakeikorako and Te Kopia area .....
2.4.1	Recent Pyroclastic Deposits .....
Hydrothermal eruption breccias .....	22
2.4.2	Pleistocene-Recent Alluvium and Hot Spring Deposits .....
Taupo Pumice Alluvium .....	24

	Hinuera Formation .....	24
	Umukuri Sinter .....	25
	Orakonui Formation .....	25
2.4.3	Pleistocene lake sediments - Huka Group .....	25
	Huka Falls Formation .....	25
	Waiora Formation .....	26
2.4.4	Quaternary igneous rocks and associated breccias .....	26
	Mangamingi Basalt .....	26
	Kakuki Basalt .....	26
	Tutua Basalt .....	28
	Parekauau Andesite .....	28
	Kaingaroa Ignimbrite .....	28
	Atiamuri Ignimbrite .....	28
	Paeroa Ignimbrite .....	28
	Te Weta Ignimbrite .....	34
	Te Kopia Ignimbrite .....	34
	Akatarewa Ignimbrite .....	35
2.4.5	Haparangi Rhyolite .....	36
2.5	Structural Features in the Active Geothermal Areas .....	36
2.5.1	Orakeikorako area .....	38
	Whakaheke Fault .....	38
	East Wainui Fault .....	38
	Faults between Whakaheke and East Wainui Faults .....	38
	Faults south of East Wainui Fault .....	38
2.5.2	Te Kopia .....	38
2.6	Geophysical Studies .....	39
2.6.1	Heat flow .....	39
2.6.2	Resistivity .....	40
2.6.3	Magnetics and gravity .....	40
2.7	Geological History of the Orakeikorako-Te Kopia Area .....	42
2.8	Orakeikorako-Te Kopia: A Single Geothermal System .....	43
 <b>Chapter 3 : Surface Hydrothermal Alteration at Te Kopia</b>		<b>45</b>
3.1	Introduction .....	45
3.2	Orakeikorako Geothermal Field: A Review .....	45
3.2.1	Orakeikorako: Discharge features/spring types .....	45
3.2.2	Orakeikorako: Surface alteration .....	47
3.3	Geology and Structure of the Te Kopia Thermal Area .....	47
3.4	Distribution of Surface Features at Te Kopia .....	48
3.5	Hydrothermal Alteration .....	63
3.5.1	Surface to near-surface alteration .....	63

	Kaolinite type .....	63
	Opal silicification .....	63
3.5.2	Alkali-chloride water alteration .....	67
	Weak clay-mordenite alteration .....	67
	Quartz-(adularia)-illite type .....	67
	Silicification .....	67
3.6	Overprinting of Quartz-(Adularia)-Illite Alteration Assemblage .....	68
3.7	Fluid Properties and Flow at Te Kopia .....	70
	Kaolinite zone .....	70
	Weak clay-mordenite zone .....	72
	Quartz-(adularia)-illite zone .....	72
3.8	Stages of Activity .....	72
3.8.1	Early alkali-chloride activity .....	73
3.8.2	Acid-sulphate activity .....	73
3.8.3	Present status .....	75
<b>Chapter 4 :</b>	<b>Fluid Chemistry and Isotopes : A review</b>	<b>76</b>
4.1	Introduction .....	76
4.2	Previous Studies .....	77
4.3	Well Discharge Chemistry .....	77
4.4	Mixing Relationships .....	82
4.5	Spring Chemistry .....	83
	Orakeikorako .....	83
	Te Kopia .....	87
4.6	Gas Chemistry .....	87
	Orakeikorako .....	87
	Te Kopia .....	90
4.7	Water Isotopes .....	90
	Orakeikorako .....	90
	Te Kopia .....	93
4.8	Gas and Sulphur Isotopes at Orakeikorako .....	93
4.9	Discussion .....	94
<b>Chapter 5 :</b>	<b>Subsurface Hydrothermal Alteration</b>	<b>96</b>
	<b>Occurrence and distribution of secondary minerals and clays</b>	
5.1	Introduction .....	96
5.2	Materials and Methods .....	97
5.3	Mineral Assemblages .....	98
5.3.1	Primary mineralogy .....	98
5.3.2	Secondary mineralogy .....	98

	Mordenite type assemblage .....	100
	Weak clay (smectite) type assemblage .....	101
	Quartz-adularia type assemblage .....	101
	Quartz silicification - A note .....	107
	Opaline silicification (sinter) .....	107
	Acid water, kaolin group type assemblage .....	107
5.4	Occurrence and Distribution of Hydrothermal Minerals .....	108
	Adularia .....	108
	Albite .....	110
	Alunite .....	113
	Analcime .....	113
	Anatase .....	113
	Anhydrite .....	113
	Calcite .....	113
	Cristobalite .....	114
	Epidote Group minerals .....	114
	Garnet (almandine) .....	119
	Hematite .....	119
	Leucoxene .....	119
	Mordenite .....	119
	Prehnite .....	122
	Pyrite .....	122
	Quartz .....	122
	Siderite .....	122
	Titanite .....	124
	Wairakite .....	124
5.5	Occurrence and Distribution of Clay minerals .....	125
	Kaolinite .....	125
	Chlorite .....	125
	Smectite .....	127
	Interstratified smectite-illite .....	127
	Illite .....	130
5.6	Discussion .....	130
<b>Chapter 6 :</b>	<b>Mineral Chemistry</b>	<b>132</b>
6.1	Introduction .....	132
6.2	Secondary minerals .....	132
6.2.1	Adularia .....	133
6.2.2	Albite .....	134
6.2.3	Calcite .....	135
6.2.4	Epidote .....	135



6.2.5	Garnet (almandine) .....	138
6.2.6	Pyrite .....	138
6.2.7	Quartz .....	140
6.2.8	Titanite .....	140
6.2.9	Zeolite minerals .....	141
	Mordenite .....	141
	Wairakite .....	141
6.2.10	Other secondary minerals .....	143
	Anatase .....	143
	Leucoxene .....	143
6.3	Clay Minerals .....	143
6.3.1	Chlorite .....	143
6.3.2	Illite .....	146
6.4	Primary Mineralogy .....	147
6.4.1	Amphibole .....	147
	Nomenclature .....	147
	Amphibole chemistry .....	147
6.4.2	Biotite .....	149
6.4.3	Hypersthene .....	152
6.4.4	Plagioclase .....	154
6.4.5	Sanidine .....	154
6.4.6	Quartz .....	154
6.4.7	Other primary minerals .....	157
	Apatite .....	157
	Magnetite .....	157
	Ilmenite .....	157
	Zircon .....	157
6.5	Summary .....	159
<b>Chapter 7 : Fluid Inclusion Study</b>		<b>160</b>
7.1	Introduction .....	160
7.2	Fluid Inclusion Microthermometry: Methodology .....	161
7.3	Fluid Inclusion Data - Orakeikorako .....	161
7.3.1	Orakeikorako Well-1 .....	162
7.3.2	Orakeikorako Well-2 .....	162
7.3.3	Orakeikorako Well-4 .....	166
7.3.4	Orakeikorako Well-6 .....	166
7.3.5	Discussion - Orakeikorako results .....	169
7.4	Fluid inclusion data - Te Kopia .....	170
7.4.1	Te Kopia Well-1 .....	173
7.4.2	Te Kopia Well-2 .....	173

7.4.3	Discussion - Te Kopia results .....	174
7.5	Surface Samples at Te Kopia .....	175
7.5.1	Results - TKS-10 .....	177
7.5.2	Results - TKS-12 .....	177
7.5.3	Discussion - Te Kopia surface samples .....	177
7.6	Raman Analysis .....	179
7.6.1	Introduction .....	179
7.6.2	Methodology .....	179
7.6.3	Composition of fluids .....	180
7.6.4	Results: Raman spectra of gases and liquids in fluid inclusions .....	180
	Raman spectrum of CO <sub>2</sub> .....	182
	Raman spectrum of other species .....	186
7.7	Discussion .....	186
 <b>Chapter 8 : Rock Chemistry</b>		<b>187</b>
<b>Ignimbrite correlation and mass transfer</b>		
8.1	Introduction .....	187
8.2	Sample Selection and Methodology .....	188
8.3	Primary Chemical Variations .....	188
8.4	Ignimbrite Correlation .....	190
8.4.1	Geochemical fingerprinting .....	190
8.4.2	Discriminant variation diagrams .....	196
8.5	Rare Earth Element Chemistry .....	202
8.5.1	Introduction .....	202
8.5.2	Results .....	202
8.5.3	REE mobility and hydrothermal alteration .....	210
8.6	Mass Transfer in the Orakeikorako-Te Kopia System .....	210
8.6.1	The isocon method .....	210
8.6.2	Results .....	213
8.6.3	Chemical change related to hydrothermal mineralogy .....	213
	Mordenite alteration zone .....	217
	Smectite alteration zone .....	217
	Quartz adularia alteration zone .....	219
8.6.4	Interpretation of mobility patterns .....	221
 <b>Chapter 9 : Thermal Evolution</b>		<b>224</b>
9.1	Introduction .....	224
9.2	Inferred Mineral Stability Temperatures .....	224
9.2.1	Orakeikorako Well-1 .....	224
9.2.2	Orakeikorako Well-2 .....	225

9.2.3	Orakeikorako Well-4 .....	225
9.2.4	Orakeikorako Well-6 .....	228
9.2.5	Te Kopia Well-1 .....	228
9.2.6	Te Kopia Well-2 .....	228
9.3	Adularia-Albite Equilibration Temperature .....	230
9.4	Illite and Chlorite Crystallinity .....	230
9.4.1	Crystallinity Index Standards .....	230
9.4.2	Clay crystallinity and basal reflection .....	232
9.5	Discussion .....	236
 <b>Chapter 10 : Chemical Evolution and Mineral-Fluid Reactions</b>		<b>241</b>
10.1	Introduction .....	241
10.2	Activity Diagrams and their Assumptions .....	242
10.3	Quartz-Adularia Alteration Assemblage .....	247
10.3.1	Temperature .....	247
10.3.2	Pressure .....	248
10.3.3	Salinity .....	248
10.3.4	Silica solubility .....	248
10.3.5	pH .....	250
10.3.6	Boiling .....	250
10.3.7	Mineral-solute equilibria .....	251
10.3.8	$f_{S_2}$ - $f_{O_2}$ and sulphide-oxide equilibria .....	252
10.3.9	Calcite overprinting .....	255
10.4	Quartz-Adularia-Illite Alteration Assemblage .....	255
10.5	Mordenite and Weak-Clay Alteration Assemblage .....	258
10.6	Kaolinite (+/- alunite) Alteration Assemblage .....	260
10.7	Summary .....	261
 <b>Chapter 11 : Concluding Discussion</b>		<b>263</b>
 <b>References</b>		<b>273</b>
 <b>Appendices</b>		<b>288</b>
 <b>Appendix A : Petrography</b>		<b>289</b>
	Definitions .....	289
	Petrologic descriptions. Orakeikorako Well 1 .....	290
	Petrologic descriptions. Orakeikorako Well 2 .....	295
	Petrologic descriptions. Orakeikorako Well 4 .....	301
	Petrologic descriptions. Orakeikorako Well 6 .....	307
	Petrologic descriptions. Te Kopia Well 1 .....	315

Petrologic descriptions. Te Kopia Well 2 .....	322
Point counting data .....	328
<b>Appendix B : Well data</b>	<b>330</b>
Downhole measurements .....	330
Well OK-1 .....	330
Well OK-2 .....	332
Well OK-4 .....	332
Well OK-6 .....	332
Well TK-1 .....	334
Well TK-2 .....	334
<b>Appendix C : Analytical methods</b>	<b>335</b>
Appendix C-1 : X-ray diffraction .....	335
Appendix C-2 : Electron microprobe .....	335
Appendix C-3 : X-ray fluorescence spectrometry .....	336
C-3a : System configuration .....	336
C-3b : Data acquisition and processing .....	337
C-3c : Sample preparation .....	337
C-3d : Accuracy and precision : X-ray fluorescence .....	339
Appendix C-4 : Neutron activation analysis .....	349
C-4a : Methodology and results .....	349
C-4b : Neutron activation results versus XRF results .....	349
Appendix C-5 : Fluid Inclusion Studies .....	356
C-5a : The Leitz-Laborlux 12HL heating-freezing stage .....	356
C-5b : Raman Analysis .....	357
<b>Appendix D : Mineral chemistry</b>	<b>359</b>
<b>Appendix E : Bulk rock chemistry</b>	<b>374</b>
<b>Appendix F : Mineral chemistry</b>	<b>382</b>
<b>Appendix G : Carbon dating</b>	<b>400</b>

## LIST OF FIGURES

Figure 1.1	Location of New Zealand	3
Figure 1.2	Location of geothermal areas in the Taupo Volcanic Zone	3
Figure 1.3	Geochemical model of a geothermal system (Hedenquist, 1986)	5
Figure 1.4	Phase diagram (sodium and potassium) in terms of ion activity at 260°C	11
Figure 2.1	Local geography of the Ngatamariki-Orakeikorako-Te Kopia-Waikite area	15
Figure 2.2	Geological sketch map of the Orakeikorako-Te Kopia area	18
Figure 2.3	Stratigraphic column of rock units identified at Orakeikorako-Te Kopia	21
Figure 2.4	Stratigraphic logs for Orakeikorako-Te Kopia drillholes	27
Figure 2.5	Crystal abundances	29
Figure 2.6a	Paeroa Ignimbrite (OK2:121.9m)	30
Figure 2.6b	Te Kopia Ignimbrite (OK6:365.8m)	30
Figure 2.6c	Te Kopia Ignimbrite (OK2:439.8m). Primary phenocrysts identified	30
Figure 2.6d	Akaterewa Ignimbrite (OK6:1219.2m)	30
Figure 2.7	Point counting data, ternary diagrams	32
Figure 2.8	Discriminant variation diagram. Zr/Y versus Yb/Hf	33
Figure 2.9	Structural map of the Orakeikorako area plus location of hot springs	37
Figure 2.10	Resistivity map	41
Figure 3.1	Location of the Te Kopia Geothermal field	46
Figure 3.2	Structural map of the Te Kopia thermal area	49
Figure 3.3	Maps of the Te Kopia Geothermal Field	50
Figure 3.4	The Te Kopia Geothermal Field, showing the extent of surface alteration	52
Figure 3.5	Paeroa Fault scarp	53
Figure 3.6	Steaming ground on the Paeroa Fault scarp	54
Figure 3.7	Acid alteration mineral assemblage	54
Figure 3.8	Acid-sulphate spring	54
Figure 3.9	Central Lakes area, Te Kopia	55
Figure 3.10	Northern Lakes area, Te Kopia	56
Figure 3.11	Two warm, steam-heated acid sulphate pools	57
Figure 3.12	Pale blue-green pool	58
Figure 3.13	Grey pool	58
Figure 3.14	Barren ground on the Paeroa Fault scarp	59
Figure 3.15	Southern Pool	59
Figure 3.16	Mud volcanoes	59
Figure 3.17	Mud geyser	61
Figure 3.18	Vent on edge of acid-sulphate pool from which mud geyser erupts	61
Figure 3.19	Swamp filled, sub-circular hydrothermal eruption crater	61
Figure 3.20	Landslide deposit at the foot of the Paeroa Fault scarp	62
Figure 3.21	A hydrothermal eruption crater in the Northern Lakes area	62
Figure 3.22	Surface hydrothermal alteration at Te Kopia	64
Figure 3.23	Silica sinter block in the Central Lakes area, Te Kopia	66
Figure 3.24	Layered sinter morphology	66
Figure 3.25	Wood enveloped in sinter	66
Figure 3.26	Block in landslide deposit on the Paeroa Fault scarp, containing quartz crystals	69
Figure 3.27	Relict 'stockwork' quartz veins	69
Figure 3.28	Abundant quartz crystals, on the ground on the Paeroa Fault scarp	69
Figure 3.29	Stages of hydrothermal activity at Te Kopia	74
Figure 4.1	Location of the Orakeikorako and Te Kopia drillholes, plus	78
Figure 4.2	Enthalpy-chloride mixing diagram for Orakeikorako water	81
Figure 4.3	Ratios of the dominant anions from Orakeikorako spring and well waters	84
Figure 4.4	Enthalpy-chloride mixing diagram for Orakeikorako waters	84
Figure 4.5	Relative Na, K, Mg-contents for Orakeikorako waters	86
Figure 4.6	Ratios of the dominant anions from Te Kopia spring and well waters	86
Figure 4.7	Plot of the ratios He, Ar and N <sub>2</sub> in geothermal gases and steam	89
Figure 4.8	Plot of $\delta^{18}\text{O}$ and $\delta\text{D}$ for Orakeikorako thermal and cold waters	92
Figure 4.9	Plot of $\delta^{18}\text{O}$ and $\delta\text{D}$ for Te Kopia thermal and cold waters	92
Figure 5.1	Alteration zones at Orakeikorako and Te Kopia	99
Figure 5.2	Quartz-adularia alteration assemblage	102
Figure 5.3	Distribution of secondary minerals and clay in OK-1	104

Figure 5.4	Distribution of secondary minerals and clay in OK-2	104
Figure 5.5	Distribution of secondary minerals and clay in OK-4	105
Figure 5.6	Distribution of secondary minerals and clay in OK-6	105
Figure 5.7	Distribution of secondary minerals and clay in TK-1	106
Figure 5.8	Distribution of secondary minerals and clay in TK-2	106
Figure 5.9	Adularia replacing plagioclase, with illite overprinting	109
Figure 5.10	X-ray diffractogram of adularia from TK-1: 884.5m	109
Figure 5.11	Plan view showing the occurrence of adularia and albite	111
Figure 5.12	Distribution of adularia in Orakeikorako-Te Kopia drillholes	112
Figure 5.13	Albite replacing plagioclase phenocrysts	112
Figure 5.14	Plan view showing the occurrence of calcite and epidote	115
Figure 5.15	Calcite replacing plagioclase phenocrysts	116
Figure 5.16	SEM electromicrograph of euhedral epidote	116
Figure 5.17	Epidote vein in Akatarewa Ignimbrite	118
Figure 5.18	Almandine garnet in vug, quartz-adularia alteration zone	120
Figure 5.19	SEM electromicrograph of mordenite filling voids	121
Figure 5.20	SEM electromicrograph of secondary quartz crystal growing into a cavity	121
Figure 5.21	SEM electromicrograph of secondary quartz veinlet	121
Figure 5.22	Plan view showing the distribution and abundance of secondary quartz	123
Figure 5.23	Replacement of hypersthene (identified by shape) by chlorite	126
Figure 5.24	SEM electromicrograph of chlorite infilling an internal cavity in hypersthene	126
Figure 5.25	Plan view showing the occurrence and distribution of smectite and illite	128
Figure 5.26	X-ray diffractogram of air-dried and glycolated smectite	129
Figure 5.27	X-ray diffractogram of air-dried and glycolated illite	129
Figure 5.28	SEM electromicrograph of illite in the groundmass of OK-4: 884.8m	131
Figure 5.29	SEM electromicrograph of illite replacing a plagioclase phenocryst	131
Figure 6.1	Ternary plot of secondary feldspar in Orakeikorako-Te Kopia drillholes	133
Figure 6.2	Compositions of calcites from the Orakeikorako-Te Kopia drillholes	136
Figure 6.3	Mol% composition of Ca, Al and (Mg+Fe <sup>2+</sup> +Fe <sup>3+</sup> ) for epidote	137
Figure 6.4	Classification of chlorites	144
Figure 6.5	Electron microprobe data. Chlorite Al(IV) and Fe/(Fe+Mg) against T <sub>bore</sub>	145
Figure 6.6	SEM electromicrograph of chlorite replacing hypersthene	146
Figure 6.7	Plot of chlorite analyses from the Te Kopia Ignimbrite	146
Figure 6.8	Compositions of calcic amphiboles from the Paeroa and Te Kopia Ignimbrites	148
Figure 6.9	Hornblende compositions for Paeroa and Te Kopia Ignimbrite; Al(IV)-Al(VI)	150
Figure 6.10	Hornblende compositions; A-site occupancy - Ti	150
Figure 6.11	Hornblende compositions; Al(IV) - Ti	150
Figure 6.12	Hornblende compositions; Al(IV) - Na(A-site)	150
Figure 6.13	Biotite from Paeroa and Te Kopia Ignimbrites; total Al versus Fe/(Fe+Mg)	153
Figure 6.14	Biotite compositions; total Al - Ti	153
Figure 6.15	Biotite compositions; Fe/(Fe+Mg) - Ti	153
Figure 6.16	Pyroxene analyses from the Te Kopia Ignimbrite	155
Figure 6.17	An-Ab-Or plot of plagioclase in the Paeroa, Te Kopia and Akatarewa Ignimbrites	156
Figure 6.18	Plot of orthoclase solid solution against anorthite content	156
Figure 7.1	Thermal profile, boiling curve for pure water, T <sub>h</sub> histograms for OK-1	164
Figure 7.2	Thermal profile, boiling curve for pure water, T <sub>h</sub> histograms for OK-2	164
Figure 7.3	Plot of T <sub>h</sub> versus T <sub>m</sub> for fluid inclusions from OK-1	165
Figure 7.4	Plot of T <sub>h</sub> versus T <sub>m</sub> for fluid inclusions from OK-2	165
Figure 7.5	Thermal profile, boiling curve for pure water, T <sub>h</sub> histograms for OK-4	165
Figure 7.6	Liquid-rich fluid inclusions from OK-6: 1141.7m	167
Figure 7.7	Liquid-rich fluid inclusions from OK-6: 1141.7m	167
Figure 7.8	Thermal profile, boiling curve for pure water, T <sub>h</sub> histograms for OK-6	168
Figure 7.9	Plot of T <sub>h</sub> versus T <sub>m</sub> for fluid inclusions from OK-4	168
Figure 7.10	Plot of T <sub>h</sub> versus T <sub>m</sub> for fluid inclusions from OK-6	168
Figure 7.11	CO <sub>2</sub> concentrations in fluid inclusions estimated by the crushing method	171
Figure 7.12	Thermal profile, boiling curve for pure water, T <sub>h</sub> histograms for TK-1	174
Figure 7.13	Thermal profile, boiling curve for pure water, T <sub>h</sub> histograms for TK-2	174
Figure 7.14	Plot of T <sub>h</sub> versus T <sub>m</sub> for fluid inclusions from TK-1 and TK-2	176
Figure 7.15	Schematic representation of the uplift of quartz from TKS-10	178
Figure 7.16	Schematic representation of the uplift of quartz from TKS-12	178
Figure 7.17	The spectrum of CO <sub>2</sub> measured in the gas phases of a 3-phase inclusion	181
Figure 7.18	CO <sub>2</sub> spectra for dry ice	183
Figure 7.19	Raman spectra of a fluid inclusion from the 50mol% CO <sub>2</sub> standard	183



Figure 7.20	Raman spectra of a fluid inclusion from the 50mol% CO <sub>2</sub> standard	184
Figure 7.21	Raman spectra of a fluid inclusion from the 25mol% CO <sub>2</sub> standard	184
Figure 7.22	Raman spectra of a fluid inclusion from the 10mol% CO <sub>2</sub> standard	185
Figure 7.23	Raman spectra of a fluid inclusion from OK-6: 1141.7m	185
Figure 8.1	Binary plots - to select immobile element pairs	192
Figure 8.2	Binary plots of Zr versus wt% P <sub>2</sub> O <sub>5</sub> , Nb and Y/Th in pumice	195
Figure 8.3	The CV (Critical Value) Test	197
Figure 8.4	Representative whole rock geochemical discriminant diagrams	198
Figure 8.5	Representative whole rock geochemical discriminant diagrams	199
Figure 8.6	Whole rock geochemical discriminant diagram of Zr/Yb versus Ta/Lu	200
Figure 8.7	Binary variation diagram - including unaltered Paeroa and Te Kopia Ignimbrites	200
Figure 8.8	Binary variation diagram - including ignimbrites from nearby systems	201
Figure 8.9	REE spidergrams for Paeroa and Te Kopia Ignimbrites	204
Figure 8.10	REE spidergrams for Te Kopia, Paeroa, A and A' Ignimbrites	206
Figure 8.11	REE spidergrams for Ignimbrite B, Te Kopia and Akatarewa Ignimbrites	208
Figure 8.12	REE spidergrams for Akatarewa, C and Kaingaroa Ignimbrites	209
Figure 8.13	Element variation diagrams; OK-2: 275.6 versus least altered Paeroa Ignimbrite	212
Figure 8.14	Isocon diagrams for representative Paeroa and Te Kopia Ignimbrites	215
Figure 8.15	Isocon diagrams for Paeroa Ignimbrites with mordenite+smectite alteration	218
Figure 8.16	Isocon diagrams for Te Kopia Ignimbrites with smectite alteration	218
Figure 8.17	Isocon diagrams for Paeroa Ignimbrites with quartz-adularia alteration	220
Figure 8.18	Isocon diagrams for Te Kopia Ignimbrites with quartz-adularia alteration	220
Figure 8.19	Binary variation diagrams; wt%CaO-Sr; wt%K <sub>2</sub> O-Rb and wt%K <sub>2</sub> O-Ba	222
Figure 9.1	Typical thermal stability range for common hydrothermal minerals in the TVZ	225
Figure 9.2	Distribution of temperature dependent secondary minerals in OK-1	226
Figure 9.3	Distribution of temperature dependent secondary minerals in OK-2	226
Figure 9.4	Distribution of temperature dependent secondary minerals in OK-4	227
Figure 9.5	Distribution of temperature dependent secondary minerals in OK-6	227
Figure 9.6	Distribution of temperature dependent secondary minerals in TK-1	229
Figure 9.7	Distribution of temperature dependent secondary minerals in TK-2	229
Figure 9.8	The K-feldspar-albite solvus	231
Figure 9.9	Crystallinity of illite and chlorite versus core depth	234
Figure 9.10	Crystallinity of illite versus measured drillhole temperatures	235
Figure 9.11	Illite and smectite basal reflections plotted against depth and bore temperature	237
Figure 9.12	Temperature profile and occurrence of clay minerals at Orakeikorako	239
Figure 9.13	Inferred temperature profile at Orakeikorako - based on hydrothermal minerals	239
Figure 9.14	Temperature profile and occurrence of clay minerals at Te Kopia	240
Figure 9.15	Inferred temperature profile at Te Kopia - based on hydrothermal minerals	240
Figure 10.1	Activity diagram for K <sup>+</sup> , Na <sup>+</sup> and H <sup>+</sup>	245
Figure 10.2	Activity diagram for K <sup>+</sup> , Ca <sup>2+</sup> and H <sup>+</sup>	246
Figure 10.3	Silica solubility	249
Figure 10.4	Mineral gas equilibria in terms of variable pH, fO <sub>2</sub> and fS <sub>2</sub>	253
Figure 10.5	Activity diagram for K <sup>+</sup> and Mg <sup>2+</sup> -bearing minerals	257
Figure 10.6	Activity diagram for the system Ca-Na-K-Al-Si	259
Figure 10.7	Dissolved silica - pH diagram showing the fluid conditions in the near surface	259
Figure 11.1	Schematic north-south cross-section through the Orakeikorako geothermal area	266
Figure 11.2	Possible hydrology of the Te Kopia thermal area, NW-SE cross-section	267
Figure 11.3	Possible hydrological model of the Orakeikorako-Te Kopia system	269

## LIST OF TABLES

Table 1.1	Typical fluid characteristics of New Zealand geothermal systems, from Hedenquist	11
Table 2.1	Hydrothermal eruptions at Orakeikorako and Te Kopia, after Lloyd (1972)	23
Table 2.2	Subsurface stratigraphy at Orakeikorako-Te Kopia, as revealed by drilling	23
Table 2.3	Average modal abundances (%) for the most widespread pyroclastic rocks	29
Table 2.4	Element ratios for the Paeroa, Te Kopia and Akatarewa Ignimbrites	33
Table 2.5	Occurrence and thickness of Te Kopia Ignimbrite	35
Table 3.1	Summary of fluid properties in the Te Kopia geothermal area	71
Table 4.1	Representative water analyses at Orakeikorako	80
Table 4.2	Representative water analyses at Te Kopia	81
Table 4.3a	Representative gas analyses, surface features at Orakeikorako and Te Kopia	88
Table 4.3b	Gas analyses from OK-2 (Orakeikorako) and TK1 (Te Kopia)	88
Table 4.4a	Isotopic composition of water and stem samples from Orakeikorako	91
Table 4.4b	Isotopic composition of water and steam samples from Te Kopia	91
Table 5.1	Common alteration products of primary minerals	102
Table 5.2	XRD data for epidote	118
Table 5.3	XRD data for mordenite	120
Table 6.1	Adularia and albite electron microprobe analyses	133
Table 6.2	Calcite electron microprobe analyses	136
Table 6.3	Epidote electron microprobe analyses	137
Table 6.4	Garnet, titanite and anatase electron microprobe analyses	139
Table 6.5	Mordenite and wairakite electron microprobe analyses	142
Table 6.6	Chlorite and illite electron microprobe analyses	142
Table 6.7	Chlorite electron microprobe analyses	145
Table 6.8	Amphibole electron microprobe analyses	148
Table 6.9	Biotite electron microprobe analyses	151
Table 6.10	Plagioclase electron microprobe analyses	155
Table 6.11	Magnetite and ilmenite electron microprobe analyses	158
Table 7.1	Fluid inclusion data from Orakeikorako	163
Table 7.2	T <sub>m</sub> and T <sub>h</sub> data from Orakeikorako	163
Table 7.3	Secondary minerals incorporated in Orakeikorako cores in fluid inclusion study	171
Table 7.4	Fluid inclusion data from Te Kopia	172
Table 7.5	T <sub>m</sub> and T <sub>h</sub> data from Te Kopia	172
Table 7.6	Secondary minerals incorporated in Te Kopia cores in fluid inclusion study	176
Table 7.7	Fluid inclusion data for Te Kopia surface samples	178
Table 7.8	Vibrational frequencies (in cm <sup>-1</sup> ) of species found in fluid inclusions	181
Table 8.1	Average composition of Paeroa and Te Kopia Ignimbrites	189
Table 8.2	Methodology for calculating mass transfer/constant volume method	191
Table 8.3	Pumice clasts in Paeroa and Te Kopia Ignimbrites	194
Table 8.4	Composition of major ignimbrite units encountered at Orakeikorako-Te Kopia	201
Table 8.5a,b	Immobile element data for the Paeroa and Te Kopia Ignimbrites	214
Table 8.6	Mass change data for the Paeroa and Te Kopia Ignimbrites	216
Table 9.1	Adularia equilibration temperatures	231
Table 9.2	Illite and chlorite crystallinity of SW standards of Warr and Rice (1993)	231
Table 9.3	Crystallinity of illite and chlorite in cores from Orakeikorako-Te Kopia drillholes	233
Table 10.1	Thermodynamic equilibrium constants for mineral-fluid reactions	243
Table 10.2	Activity ratios estimated from well chemistry data	245
Table 11.1	Evolution of the Orakeikorako-Te Kopia geothermal system	264



## Acknowledgements

It is with the greatest respect that I acknowledge the contribution that Assoc. Prof. Patrick (Pat) R. L. Browne has made, as my main supervisor, to the completion of this thesis. Pat has been a ready source of information, ideas and encouragement and I thank him for the time he has freely given. The 'lazy reader' will also benefit from the suggestions he has made to improve this thesis. My second supervisor has been Professor P. M. Black. I do appreciate her assistance, mostly 'behind the scenes', but also in helping to organise my JSPS scholarship to Misasa, Okayama University for 1994-1995, and for reading this thesis.

Many other people have helped me overcome obstacles (theoretical, practical and sometimes imagined) during the last four years, to whom I owe my sincere thanks.

- The staff of the Geothermal Institute, University of Auckland; academic, technical and administrative, but in particular Drs. Stuart Simmons and Keith Nicholson.
- Dr. Ritchie Sims (Electron microprobe); John Wilmshurst and Dr. Robin Parker (XRF) and Sue Courtney (XRD and SEM); Assoc. Prof. Kerry Rodgers (Laser Raman); all staff of the Geology Dept., University of Auckland.
- Dr. John Seakins (Laser Raman); Chemistry Dept.
- Dr. Phil Kyle (Neutron Activation Analysis), Dept. of Earth Sciences, New Mexico Institute of Technology, Socorro, New Mexico.
- Staff of the Institute of Nuclear and Geological Sciences (IGNS) Wairakei; particularly Dr. C.P. Wood and L.E. Klyen who made unpublished information available to me.
- Most of the thin sections of the Orakeikorako and Te Kopia drillholes described in this thesis were kindly loaned by the Geological Survey (DSIR); now IGNS. Thanks to Jan Hanbury-Sparrow (Curator, Petrology Collection).
- Dr. Alan Hogg (Waikato University) and Dr. Roger Spakes (IGNS, Lower Hutt) conducted  $C^{14}$  dating on the wood sample enveloped in sinter (Te Kopia).
- The Waikato Regional Council provided some funding to help with field expenses- I would like to thank Dr. Jim McLeod for organising this support.
- The Geothermal Institute provided the opportunity to tutor and demonstrate for four years of Diploma students. It was an enjoyable experience. I learnt from it and I trust the students did too. The Geothermal Institute, and in particular the common room, would be a duller place without the graduate students. Some have gone during the 1990-1994 period, whilst others have just arrived but Nenny Saptadji, Juliet Newson, Zhang Lan, Lisa Koenig, Huang Yicun, Djoko Suranto, Steve Torrens, Markos Melaku and Chen Song are the friends that the stories will be about.
- To my family, thanks for everything you have done to help.
- Finally, to Lisa, whom I love very much. It would not have been possible to complete this work without her immeasurable support. Thank you so very very much Lisa.