Logarithmic mean: Chen's approximation or explicit solution?

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Abstract

An explicit solution has been obtained for the logarithmic mean temperature difference method of heat exchanger calculation by making use of the Lambert W-function. The results might be of use where an explicit solution involving the logarithmic mean is required.

Chen's Approximation

An approximation to the logarithmic mean (LMTD) between two numbers θ_1 and θ_2 attributed to Chen (1987) is given in Eq. (1).

$$LMTD = \frac{\theta_1 - \theta_2}{\ln\left(\frac{\theta_1}{\theta_2}\right)} \approx \left\{\theta_1, \theta_2, \left(\frac{\theta_1 + \theta_2}{2}\right)\right\}^{\frac{1}{3}}$$
 Eq. 1

Eq. (1) has been widely used in optimization models for heat exchanger network and other engineering equipment (see, for example, Yee et al. 1990, Yee & Grossman, 1990, Amarger et al, 1992, Lewin, 1998, Adjiman et al, 2000, Jackson & Grossmann, 2001, Davis & Sandall, 2003, Ponce-Ortega et al, 2008, Gabriel, et al., 2016, Bongartz & Mitsos, 2017, Pavao et al., 2017). The Chen approximation slightly overestimates the area requirement, but it avoids numerical problems associated with the logarithmic term, and it also has the important advantage that when either θ_1 or θ_2 equals zero the driving force will Accepted for publication in Computers and Chemical Engineering, 6 October 2018 be approximated to be zero (Yee et al. 1990, Yee & Grossmann, 1990, Lewin, 1998, Floudas et al., 1999, Adjiman et al., 2000, Davis & Sandall, 2003).

Lambert W-function

The earliest mention of the problem stated in Eq. 2 is attributed to Euler (1779), but Euler himself credited Lambert (1758) with it (see also Corless et al. 1996, Hayes 2005). Corless et al. (1996) gave a detailed analysis of the Lambert W-function while Barry et al. (2000) gave analytical approximations for it. Hayes (2005) and Stewart (2005) gave simplified accounts of the properties of the Lambert W-function. Others, including Keady (1998) who applied the Lambert W-function to the Colebrook-White equation, Valluri et al. (2000) who discussed its possible applications in physics, and Golicnik (2012) applied the function to enzyme-catalysed biochemical reactions, while Disney & Warburton (2012) applied it to certain economic problems. Hayes (2005) suggests that scientific calculators should have a built in Lambert key, and on the *American Scientist* Website, Hayes (2005) asked: "Should Lambert W be added to the canon of standard textbook functions?" Stewart (2005) also suggested that the Lambert W function should be included in the mathematical curricula.

In Eq. 2, it is easy to calculate y when the value of x is given.

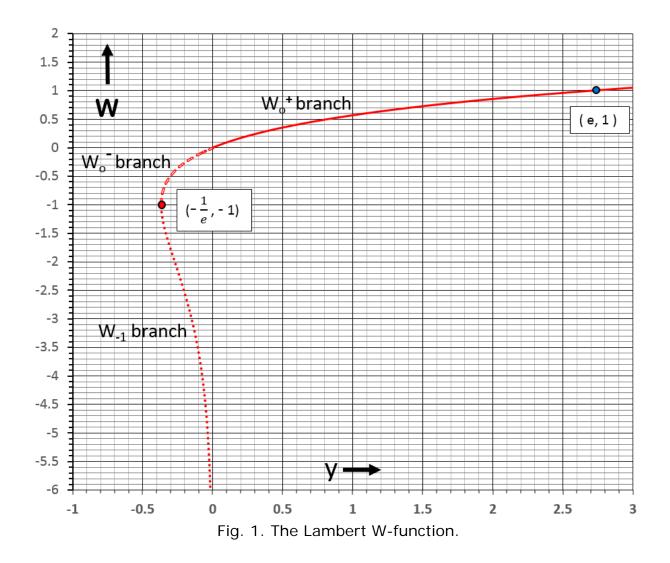
$$xe^x = y$$
 Eq. 2

However, in the inverse case, i.e. given y, the value of x is not readily obtained. This is where the Lambert W-function comes in because it is the inverse of Eq. 2, i.e. if y is the input, the Lambert W-function gives x = W(y) such that Eq. 2 is satisfied. Accepted for publication in Computers and Chemical Engineering, 6 October 2018 For example, with reference to Fig. 1, for y=e, W(e) gives x=W(y)=1 because e = 1*e¹, and this is plotted as (e, 1); and for $y = \left(-\frac{1}{e}\right)$, $W\left(-\frac{1}{e}\right)$ gives x=W(y)=-1 because $-1*e^{-1} = -e^{-1}$, and this is plotted as $\left(-\frac{1}{e}, -1\right)$.

The definition of W(y) may be written as

$$x = W(y) = W(xe^x)$$
 Eq. 3

The inverse W function given by Eq. 3 has two values of W when y is between $-\frac{1}{e}$ and 0. When y is greater than 0, W is single valued (Hayes, 2005).



The curve may be divided into three regions (Golicnik, 2012, Barry et al., 2000) and the branches are:

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- 1. Region 1: y>0, the W_0^+ branch;
- 2. Region 2: $-e^{-1} < y < 0$, and 0 > W > -1, the W_o⁻ branch;
- 3. Region 3: $-e^{-1} < y < 0$, W < -1, the W₋₁ branch.

Application of the Lambert W-function to the Logarithmic Mean

Writing the heat exchanger equation as

$$Q = UA \frac{\theta_1 - \theta_2}{\ln(\frac{\theta_1}{\theta_2})}$$
 Eq. 4

Dividing by θ_2 and re-arranging

$$\frac{Q}{UA\theta_2} = \frac{\frac{\theta_1}{\theta_2} - 1}{ln\frac{\theta_1}{\theta_2}}$$
 Eq. 5

Consider θ_2 as known and θ_1 the unknown to be solved,

Let
$$\frac{\theta_1}{\theta_2} = \beta$$
 Eq. 6

Let
$$\frac{Q}{UA} = K$$
, such that $\frac{K}{\theta_2} = \frac{\beta - 1}{\ln \beta}$ Eq. 7

Thus,
$$\ln \beta = \frac{\theta_2}{\kappa} (\beta - 1) = \frac{\theta_2}{\kappa} \beta - \frac{\theta_2}{\kappa}$$
 Eq. 8

Giving
$$\beta = e^{\left(\frac{\theta_2}{K}\beta - \frac{\theta_2}{K}\right)} = \frac{e^{\frac{\theta_2}{K}\beta}}{e^{\frac{\theta_2}{K}}}$$
 Eq. 9

Eq. 9 may be re-cast as

$$\beta e^{-\frac{\theta_2 \beta}{K}} = e^{-\frac{\theta_2}{K}}$$
 Eq. 10

Multiplying both sides by $-\frac{\theta_2}{K}$

$$-\frac{\theta_2}{\kappa}\beta e^{-\frac{\theta_2}{\kappa}\beta} = -\frac{\theta_2}{\kappa}e^{-\frac{\theta_2}{\kappa}}$$
Eq. 11

Accepted for publication in Computers and Chemical Engineering, 6 October 2018 In Eq. 11, θ_2 is known, while β the unknown to be solved appears on the lefthand-side of the equation only.

This is where the Lambert W function comes in. Taking the W of both sides of Eq. 11, the LHS is of the form xe^x , where $x = -\frac{\theta_2}{\kappa}\beta$ according to Eq. 3.

Thus, Eq. 11 becomes

$$-\frac{\theta_2}{\kappa}\beta = W\left(-\frac{\theta_2}{\kappa}e^{-\frac{\theta_2}{\kappa}}\right)$$
 Eq. 12

All terms on the RHS of Eq. 12 that are within the W function are known and hence $W\left(-\frac{\theta_2}{K}e^{-\frac{\theta_2}{K}}\right)$ can be evaluated. In fact, by inspection with reference to Eq. 3, it should give a numerical value of $-\frac{\theta_2}{K}$. However, from Fig. 1, W of a negative number for $-e^{-1} < y < 0$ will yield two roots, and the value of $-\frac{\theta_2}{K}$ is expected to be one of them. The two roots are obtained as one each from the W_0^- branch and the W₋₁ branch. Thus β can be evaluated.

Example 1 (for the general case when $\theta_1 \neq \theta_2$)

To illustrate the application of Eq. 12, using the same problem considered in Patterson (1984):

- $\theta_2 = (50-30) = 20^{\circ}C;$ $\theta_1 = (125-t_1)^{\circ}C;$ $\beta = \theta_1 / \theta_2$
- Q = 7500kW; $UA = 175 kW/^{\circ}C$
- $K = Q/UA = 7500/175 = 42.857^{\circ}C$
- $\theta_2 / K = 20/42.857 = 0.4667$

Thus, if β is solved, θ_1 and t_1 can be readily evaluated.

Accepted for publication in Computers and Chemical Engineering, 6 October 2018 Applying Eq. 12, the LHS gives -0.4667 β , and the RHS gives W(-0.4667 $e^{-0.4667}$), i.e. W(-0.2927). From Figure 1, W(-0.2927) has two roots, one from the W₀⁻ branch and one from the W₋₁ branch.

The exact values of the two roots may be obtained using a standard mathematical software package such as Wolfram MathWorld, Matlab, Macsyma, Maple, Mathematica etc. (Hayes 2005). Note that the Lambert W-function is also referred to as the Product Log function in some software packages. Barry et al. (2000) and Golicnik (2012), among others, have provided equations for the evaluation of W, and tabulated values are also given in Disney & Warburton (2012).

W(-0.2927) gives -0.46 from the W₀⁻ branch (which is $-\frac{\theta_2}{K}$ = -0.4667 as

discussed earlier) and -1.84 from the W₋₁ branch. Thus

$$-0.4667\beta = -0.46$$
; and $-0.4667\beta = -1.84$

The first solution is 'trivial' as this occurs when $\beta = 1$, but $\theta_1 \neq \theta_2$, and hence this root can be discarded. The second solution gives $\beta = 3.94$, resulting in $\theta_1 = 78.9$ and $t_1 = 46.1^{\circ}$ C. As expected, this solution is exactly the same as that obtained by iteration using the actual LMTD.

Example 2 (for the case when $\theta_1 = \theta_2$)

Basically the Paterson (1984) case but setting θ_2 to be equal to K.

 $\theta_2 = (72.86-30) = 42.86^{\circ}C;$ $\theta_1 = (125-t_1)^{\circ}C;$ $\beta = \theta_1 / \theta_2$ Q = 7500kW; $UA = 175 \ kW/^{\circ}C$

 $K = Q/UA = 7500/175 = 42.857^{\circ}C$

Accepted for publication in Computers and Chemical Engineering, 6 October 2018 $\theta_2 / K = 42.86/42.857 \approx 1.$

Applying Eq. 12, the LHS gives – β , and the RHS gives W(-e⁻¹). From Figure 1, this occurs at the junction of W₀⁻ and W₋₁, i.e. there is only one root, and W(-e⁻¹) = -1, i.e. β =1 giving $\theta_1 = \theta_2$.

Discussion

The two examples shown above are based on the case considered by Paterson(1984) and Chen(1987) using the transformed equation given in Eq. 12. Eq. 12 is equivalent to Eq. 4, but it allows for the exact evaluation θ_2 without a trial-and-error process which is necessary when using Eq. 4 because of the appearance of θ_2 in two places on the right-hand-side of the equation, inside and outside of the logarithmic term. In the case where Q (or A) is the only unknown, Q may be evaluated from an energy balance, and Eq. 4 will provide an explicit solution.

The above examples considered θ_2 as the unknown. If θ_1 is the unknown instead and is required to be solved, it is simply a matter of interchanging θ_1 and θ_2 in Eq. 4, 5, and 6, and the solution will follow immediately.

It is also noted that in the case where $\theta_1 = \theta_2$ as illustrated in Example 2, Eq. 4 will become $Q = UA \frac{\theta_1 - \theta_2}{\ln(\frac{\theta_1}{\theta_2})} = UA\theta_1 = UA\theta_2$.

Conclusion

An explicit solution has been shown for heat exchanger calculations using the logarithmic mean temperature method by employing the Lambert W-function due originally to Lambert (1758) and Euler (1779). It remains to be seen if it

Accepted for publication in Computers and Chemical Engineering, 6 October 2018 will replace the use of approximations for the logarithmic term in the research

on optimization such as those referenced in this note.

Declaration of Interest

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<u>References</u>

Adjiman, C.S., Androulakis, I.P., Floudas, C.A., 2000, Global optimization of mixedinteger nonlinear problems, *AIChE JI.*, **46**, 1769-1797.

Amarger, R.J., Biegler, L.T., Grossmann, I.E., (1992) An automated modelling and reformulation systems for design optimization, Computers and Chemical Engineering, 16, 623-636.

Barry, D.A., Parlange, J.-Y., Li, L., Prommer, H., Cunningham, C.J., Stagnitti, F. (2000), Analytical approximations for real values of the Lambert W-function, Mathematics and Computers in Simulation, 53, 95-103.

Bongartz, D., Mitsos, A., 2017, Deterministic global optimization of process flowsheets in a reduced space using McCormick relaxations, J. Global Optimization, 69, 761-796.

Chen, J. J. J., 1987, Letter to the editor: comments on improvement on a replacement for the logarithmic mean. Chemical Engineering Science, 42, 2488-2489.

Corless, R.M., Gonnet, G.H., Hare, D.E.G., Jeffrey, D.J., Knuth, D.E., 1996, On the lambert W function, Advances in Computational Mathematics, 5, 329-359.

Davis, R.A., Sandall, O.C., 2003, A Simple Analysis For gas separation membrane experiments, Chemical Engineering Education, 74-80, Winter 2003.

Disney, S.M., Warburton, R.D.H., 2012, On the Lambert W function: Economic order quantity applications and pedagogical considerations, Intl. J. Production Economics, 40, 754-764.

Euler, L., 1779, De serie Lambertina plurimisque eius insignibus proprietatibus, Leonhardi Euleri Opera Omnia, Ser. 1, Opera Mathematica 6 (1921) [orig. date 1779] 350-369.

Floudas, C. A., Pardalos, P.M., Adjiman, C.S., Esposito, W.R., Gumus, Z.H., Harding, S.T., Klepeis, J.L., Meyer, C.A., and Schweiger, C.A., 1999, Handbook of Test Problems in Local and Global Optimization, Kluwer Academic Press.

Gabriel, K.J., El-Halwagi, M.M., Linke, P., 2016, Optimization across the Water-Energy Nexus for Integrating Heat, Power, and Water for Industrial Processes, Coupled with Hybrid Thermal-Membrane Desalination, Industrial & Engineering Chemistry Research, 55, 3442-3466.

Golicnik, M., 2012, On the Lambert W function and its utility in biochemical kinetics, Biochemical Engineering Journal, 63, 116-123.

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Hayes, B., 2005, Why W? American Scientist, 93, 104-108. See also <u>https://www.americanscientist.org/article/why-w</u> "Why W? "Should Lambert W be added to the canon of standard textbook functions?" (Downloaded 18 September 2018)

Jackson, JR; Grossmann, IE., 2001, A disjunctive programming approach for the optimal design of reactive distillation columns, Computers and Chemical Engineering, 25, 1661-1673.

Keady, G., 1998, Colebrook-White formula for pipe flows, J. Hydraulic Engineering, 124, 96-97.

Lambert, L.J.H., 1758, Observationes variae in mathesin puram, Acta Helvetica, physico-mathematicoanatomico-botanico-medica 3, Basel (1758) 128-168.

Lewin, DR, 1998, A generalized method for HEN synthesis using stochastic optimization - II. The synthesis of cost-optimal networks, Computers and Chemical Engineering, 22, 1387-1405.

Paterson, W. R., 1984, A replacement for the logarithmic mean. Chemical Engineering Science, 39, 1635-1636.

Pavao, L.V., Borba Costa, C.B., Ravagnani, M. A. S. S., 2017, Heat Exchanger Network Synthesis without stream splits using parallelized and simplified simulated Annealing and Particle Swarm Optimization, Chemical Engineering Science, 158, 96-107.

Ponce-Ortega, J. M., Jimenez-Gutierrez, A., Grossmann, I.E., 2008, Optimal synthesis of heat exchanger networks involving isothermal process streams, Computers and Chemical Engineering, 32, 1918-1942.

Stewart, S., 2005, A new elementary function for our curricula? Australian Senior Mathematics Journal, 19, 8-26.

Yee, T.F., Grossmann, I.E., Kranja, Z., 1990, Simultaneous-optimization models for heat integration. 1. Area and energy targeting and modelling of multi-stream exchangers, Computers and Chemical Engineering, 14, 1151-1164.

Yee, T.F., Grossmann, I.E., 1990, Simultaneous-optimization models for heat integration. 2. Heat-exchanger networks synthesis, Computers and Chemical Engineering, 14, 1165-1184.