# ChemBioChem

Combining Chemistry and Biology



European Chemical Societies Publishing



## **Accepted Article**

Title: Carbon monoxide is an inhibitor of HIF prolyl hydroxylase domain 2

Authors: Naasson M. Mbenza, Nawal Nasarudin, Praveen G. Vadakkedath, Kamal Patel, Ahmad Zaidi Ismail, Muhammad Hanif, Leonard James Wright, Vijayalekshmi Sarojini, Christian G. Hartinger, and Ivanhoe K. H. Leung

This manuscript has been accepted after peer review and appears as an Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). This work is currently citable by using the Digital Object Identifier (DOI) given below. The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: ChemBioChem 10.1002/cbic.202100181

Link to VoR: https://doi.org/10.1002/cbic.202100181

WILEY-VCH

# COMMUNICATION

# Carbon monoxide is an inhibitor of HIF prolyl hydroxylase domain 2

Naasson M. Mbenza,<sup>[a, b]</sup> Nawal Nasarudin,<sup>[a]</sup> Praveen G. Vadakkedath,<sup>[a, c]</sup> Kamal Patel,<sup>[a]</sup> A. Z. Ismail,<sup>†[a]</sup> Muhammad Hanif,<sup>[a, d]</sup> L. James Wright,<sup>[a]</sup> Vijayalekshmi Sarojini,<sup>[a, c]</sup> Christian G. Hartinger,<sup>[a, d]</sup> and Ivanhoe K. H. Leung\* <sup>[a, d, e, f]</sup>

**Abstract:** Hypoxia-inducible factor prolyl hydroxylase domain 2 (PHD2) is an important oxygen sensor in animals. By using the CO-releasing molecule-2 (CORM-2) as an *in situ* CO donor, we demonstrate that CO is an inhibitor of PHD2. This report provides further evidence about the emerging role of CO in oxygen sensing and homeostasis.

Carbon monoxide (CO) is a colourless and odourless gaseous molecule. Although most commonly known as an acute toxin that outcompetes molecular oxygen for the binding of haemoglobin at high concentrations,<sup>[1]</sup> CO is produced endogenously in humans by haem oxygenases as a co-product of haem catabolism.<sup>[2–5]</sup> At physiologically relevant concentrations, CO is a signalling molecule that is involved in a number of important physiological functions.<sup>[5–9]</sup> It has been shown to mediate signalling processes in the brain,<sup>[10,11]</sup> induce vasorelaxation,<sup>[12,13]</sup> suppress inflammation,<sup>[14]</sup> modulate mitochondrial functions,<sup>[15–17]</sup> and affect cell proliferation and apoptosis in tissue- and cell-specific manners.<sup>[18–22]</sup>

One of the most intriguing aspects of CO is its relationship with cellular oxygen homeostasis. Not only is the endogenous

[a]	Dr N. M. Mbenza, N. Nasarudin, Dr P. G. Vadakkedath, Dr K. Patel,
• •	Dr A. Z. Ismail, Dr M. Hanif, Prof. L. J. Wright, A/Prof. V. Sarojini,
	Prof. C. G. Hartinger, Dr I. K. H. Leung
	School of Chemical Sciences
	The University of Auckland
	Private Bag 92019, Victoria Street West
	Auckland 1142 (New Zealand)
[b]	Dr N. M. Mbenza
• •	School of Biological Sciences
	Victoria University of Wellington
	PO Box 600, Wellington 6140 (New Zealand)
[c]	Dr P. G. Vadakkedath, A/Prof. V. Sarojini
	The MacDiarmid Institute for Advanced Materials and
	Nanotechnology
	Victoria University of Wellington
	PO Box 600, Wellington 6140 (New Zealand)
[d]	Dr M. Hanif, Prof. C. G. Hartinger, Dr I. K. H. Leung
	Maurice Wilkins Centre for Molecular Biodiscovery
	The University of Auckland
	Private Bag 92019, Victoria Street West
	Auckland 1142 (New Zealand)
[e]	Dr I. K. H. Leung
• •	School of Chemistry
	The University of Melbourne
	Parkville, VIC 3010 (Australia)
	Email: ivanhoe.leung@unimelb.edu.au
[f]	Dr I. K. H. Leung
	Bio21 Molecular Science and Biotechnology Institute
	The University of Melbourne
	Parkville, VIC 3010 (Australia)
†	Current address: Department of Chemistry, Universiti Putra
•	Malaysia, 43400, Serdang, Selangor (Malaysia)
	Supporting information for this article is given via a link at the end of

the document.

production of CO oxygen-dependent (as haem oxygenases require molecular oxygen as co-substrate),<sup>[2–5]</sup> there is also increasing evidence that shows CO may play critical roles in cellular oxygen sensing and signalling.<sup>[23,24]</sup> Of particular interest is the observation that, in normoxia, physiologically-relevant concentrations of CO may activate and stabilise hypoxia-inducible factor  $\alpha$  (HIF $\alpha$ ).<sup>[18,25]</sup> In hypoxia, in contrast, it has been shown that CO may actually promote the degradation of HIF $\alpha$ .<sup>[17,26,27]</sup> HIF $\alpha$  is a master regulator of hypoxic responses to maintain oxygen homeostasis.<sup>[28–34]</sup> During hypoxia, HIF $\alpha$  dimerises with HIF $\beta$  and other co-activators to transcriptionally activate a number of genes, such as those that are involved in erythropoiesis and angiogenesis, to counteract the effect of low oxygenation.<sup>[28–34]</sup>

The activity of HIF is regulated by a complex network of multiple intertwined pathways. One of these pathways regulates HIFa at the level of protein stability.<sup>[28-34]</sup> HIFa and HIFB are constitutively produced regardless of cellular oxygen concentrations. HIFa is subjected to oxygen-dependent proteolytic degradation. This process is regulated by HIF prolyl hydroxylase domain-containing proteins (PHDs), which, in humans, exist in three isoforms.<sup>[28-34]</sup> PHDs belong to the Fe(II) and 2-oxoglutarate (2OG)-dependent oxygenase superfamily.[28-<sup>34]</sup> Members of this superfamily typically require Fe(II), 2OG, molecular oxygen and ascorbate for optimal activity.[35-38] PHDs catalyse the hydroxylation of HIFa in the presence of oxygen.<sup>[28-</sup> <sup>34]</sup> The resultant hydroxyl group(s) on HIFα acts as a signal for its degradation via the ubiquitin-proteasome pathway through binding to the von Hippel-Lindau protein.[28-34]

In human HIF1 $\alpha$ , two prolyl residues, Pro-402 and Pro-564, are involved in this pathway.<sup>[28–34]</sup> The two hydroxylation sites are termed the N-terminal oxygen-dependent degradation domain (NODD, where Pro-402 is present) and C-terminal oxygen-dependent degradation domain (CODD, where Pro-564 is present). It has been shown that the hydroxylation of either CODD or NODD may lead to the degradation of HIF1 $\alpha$ .<sup>[39,40]</sup> Although NODD hydroxylation is more sensitive to changes in oxygen concentration than CODD hydroxylation.<sup>[41]</sup> kinetics and binding studies showed that CODD is the preferred substrate and is a stronger binder of PHDs than NODD, which is also supported by structural studies.<sup>[42–45]</sup>

Given the central position that PHDs play in regulating hypoxic response, the modulation of PHDs by endogenously produced compounds is of significant interest from both therapeutic and basic biology perspectives. CO, in general, has a high affinity for ferrous Fe(II). For example, the affinity of CO to the reduced iron-haem in haemoglobin is about 220 times stronger than that of molecular oxygen.<sup>[46]</sup> In addition, it has also been shown that CO could bind to a range of metalloproteins, including iron-dependent enzymes such as iron and nickel iron hydrogenases,<sup>[47-51]</sup> as well as copper-dependent proteins including tyrosinases and haemocyanins.<sup>[52-54]</sup> Given that PHDs

#### COMMUNICATION

are Fe(II)-dependent metalloenzymes, we speculate that CO may inhibit PHDs by competing with molecular oxygen for binding at the active site Fe(II). Herein, by using the catalytic domain of PHD2 (PHD2<sub>181-426</sub>), the PHD isoform that is considered to be the most important oxygen sensor,<sup>[55–57]</sup> as a model system, we provide evidence that demonstrates CO is an inhibitor of PHDs.

As CO is a gaseous molecule, CO-releasing molecule-2 (CORM-2; [RuCl<sub>2</sub>(CO)<sub>3</sub>]<sub>2</sub>) was used as an in situ CO donor. The use of CORMs (including CORM-2) as CO donors in biological systems is well established.<sup>[58-60]</sup> Nonetheless, experiments were first conducted to investigate the CO-release kinetics under our reaction condition. In our reactions, CORM-2 was first dissolved in dimethyl sulfoxide (DMSO), which was then immediately added to the aqueous buffer containing the enzyme mixture. In agreement with a previous study conducted by Motterlini and coworkers,<sup>[59]</sup> we found that CORM-2, when freshly dissolved in DMSO, decomposed mainly into fac-[RuCl<sub>2</sub>(CO)<sub>3</sub>(DMSO)] and cis, cis, trans-[RuCl<sub>2</sub>(CO)<sub>2</sub>(DMSO)<sub>2</sub>], indicating that there was minimal liberation of CO (at least) in the first few minutes (Figure S1). We then measured the kinetics and the amount of CO that was released after the CORM-2/DMSO stock solution was added to aqueous buffer. A CO-trapping assay that monitors the conversion of deoxymyoglobin to carboxymyoglobin by ultraviolet/visible (UV/vis) absorption spectroscopy was applied.<sup>[61]</sup> 2OG and ascorbate were also added to the aqueous solution to mimic the condition for PHD2-catalysed reactions. Upon addition of the freshly prepared CORM-2/DMSO solution (to give 100 µM CORM-2 final concentration) to a buffer solution containing 50 µM deoxymyoglobin, full conversion of deoxymyoglobin to carboxymyoglobin was observed in the first UV/vis spectral measurement (ten seconds after mixing) (Figure S2), indicating liberation of CO from CORM-2 in aqueous buffer. Titration of CORM-2 to deoxymyoglobin (50 µM) showed that all myoglobin was saturated with CO at ~1:1 CORM-2 to myoglobin ratio (Figure S3), indicating that, for every CORM-2, roughly one CO was released. This is in line with a previous study that showed around 0.7 moles of CO was liberated from each mole of CORM-2.<sup>[59]</sup> The difference is likely due to handling of the CORM-2 stock solution and buffer conditions. These experiments showed that CORM-2 is an efficient source of CO in aqueous buffer.

Another potential concern about the use of CORM-2 to study PHD2-catalysed reactions is the possibility for decomposed or non-decomposed CORM-2 (rather than CO) to inhibit PHD2. This concern is particularly valid for metalloenzymes such as PHD2 as it has been reported that some transition metals may inhibit the enzyme by displacing the active site Fe(II).[62,63] As  $RuCl_3^{[17,64,65]}$  and  $RuCl_2(DMSO)_4^{[64,66,67]}$  have been used as negative controls for biological experiments using CORMs, the ability of these compounds to inhibit PHD2 was investigated. By using a matrix assisted laser desorption ionisation-time of flight mass spectrometry (MALDI-TOF MS)-based assay that peptide hydroxylation, and measures 19-mer HIF1α CODD/NODD peptides as substrates (see Table 1 for peptide sequences),<sup>[24]</sup> we found that RuCl<sub>3</sub> is an inhibitor of PHD2 under the standard assay condition with 1 µM PHD2 and 10 µM Fe(II) (Figure S4). However, this effect can be alleviated by supplementing the enzymatic reaction mixture with excess Fe(II) (for example, >200 µM; Figure S4). Similarly, under the same reaction condition with 200 µM Fe(II), RuCl<sub>2</sub>(DMSO)<sub>4</sub> did not inhibit PHD2 at up to 1 mM concentration (Figure S5).

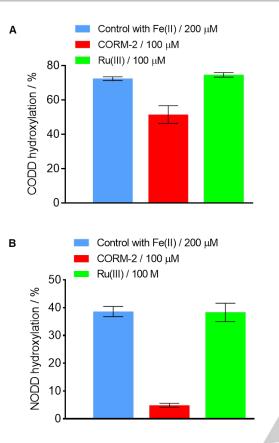
We then tested the inhibition of PHD2 by CORM-2. Under our optimised reaction conditions with 1  $\mu$ M PHD2 and excess Fe(II) (200 or 400 µM), we found that CORM-2 reduced the hydroxylation of CODD and NODD (Figure 1 and Figure S6). As PHD2-catalysed CODD/NODD hydroxylations were not affected by RuCl<sub>3</sub> or RuCl<sub>2</sub>(DMSO)<sub>4</sub> under these conditions, our results indicated that CO was likely the species that inhibits PHD2. To further confirm the inhibition of PHD2 by CO, competition experiments were performed with deoxymyoglobin. This is because deoxymyoglobin is a good CO binder and it will provide competition with PHD2 for the binding of CO. We found that the addition of 10 µM deoxymyoglobin restores the activity of PHD2 in the presence of 75 µM CORM-2 (Figure S7). As one myoglobin could bind one molecule of CO, the additional protection effect is intriguing. We therefore repeated the experiments in the absence of CORM-2. We found that the addition of deoxymyoglobin led to a slight increase in PHD2-catalysed hydroxylation reactions (Figure S7). As myoglobin is known to react with hydrogen peroxide.[68] the slight increase in PHD2 activity could be due to myoglobin removing the hydrogen peroxide that was produced as a result of PHD2-catalysed reactions. Such effects have been observed with catalase in 20G-dependent oxygenase-catalysed reactions.<sup>[69]</sup> Nonetheless, even taking this slight increase of PHD2 activity into account, our results still point towards CO as the main species that inhibits PHD2.

There are reports that some proteins may form adducts with CORM-2 through interactions between surface-exposed histidine residues and the Ru(II) centre of CORM-2.[70,71] As such, it is possible that CORM-2 may react covalently with PHD2 and deactivates PHD2 in a CO-independent manner. However, it was also suggested that the addition of excess imidazole or serum albumin may suppress the formation of the CORM-2-protein covalent adducts as imidazole and serum albumin may act as competing molecular targets to react with CORM-2.[71] We therefore repeated the inhibition experiments with CORM-2 (150 µM) in the presence of 1 mM imidazole or bovine serum albumin (BSA). Our results showed that the inhibition of PHD2 by CORM-2 was not affected by the presence of imidazole (Figure S8). We observed a slight increase of PHD2 activity in the presence of BSA (although the difference could be within error; Figure S8). Hence, although we cannot rule out that some CORM-2 may react covalently with PHD2, our results strongly indicate that the COdependent pathway is the predominant inhibition mechanism.

To further evaluate the effect of CO on PHD2 catalytic activity, we also performed PHD2 inhibition experiments by directly bubbling CO into a solution containing PHD2-substrate mixture. Our results showed that the PHD2-catalysed CODD hydroxylation was fully inhibited after bubbling CO through the solution for 10 seconds (Figure S9). However, as the bubbling of CO through the solution may also displace molecular oxygen from the buffer, careful interpretation of the data is required. Binding studies using native mass spectrometry were also attempted as previous studies have shown that it was possible to observe the binding of myoglobin to CO.<sup>[72]</sup> However, despite numerous attempts, we were unable to directly observe the PHD2-Fe(II)-CO complex.

Accepted Manuscri

### COMMUNICATION



**Figure 1.** CORM-2 inhibits the hydroxylation of (A) CODD and (B) NODD to different extents. The reaction mixtures contained 1  $\mu$ M PHD2, 200  $\mu$ M Fe(II), 60  $\mu$ M 2OG, 50  $\mu$ M CODD/NODD, 100  $\mu$ M CORM-2 or RuCl<sub>3</sub> (where applicable), 500  $\mu$ M ascorbate in 50 mM Tris (pH 7.5). The reaction mixtures were incubated at 37 °C and the concentration of hydroxylated CODD/NODD was measured after 30 minutes of incubation. The results are expressed as mean ± standard error of the mean (n = 3).

Comparing the hydroxylation of CODD and NODD, we found that CO has a much stronger inhibitory effect on NODD hydroxylation than CODD hydroxylation (Figure 1). At 100 µM CORM-2 concentration, ~80% reduction of NODD hydroxylation was observed compared to only ~30% reduction of CODD hydroxylation under similar conditions. In order to quantify this effect, inhibition constants  $(IC_{50})$  were determined. The IC<sub>50</sub> values of CORM-2 when NODD and CODD were used as substrates were found to be 34.2  $\pm$  1.3  $\mu M$  and 142.5  $\pm$  1.8  $\mu M,$ respectively (Table 1 and Figure S10). This is an interesting observation, as our results showed that different substrates may modulate the inhibition potency of CO. CODD is a stronger binder than NODD, which is reflected by the binding constants ( $K_D$ values) (~14 µM and ~85 µM for CODD and NODD, respectively)<sup>[45]</sup> and, to a certain extent, the Michaelis constants (K\_M values) for these two compounds (5.2  $\pm$  1.3  $\mu M$  and 107.1  $\pm$ 29.7 µM, respectively; Table 1 and Figure S11). In order to further study the effect of CO modulation of PHD2 activity towards different substrates, additional PHD2 substrates were explored. Previous studies showed that the N-terminal side of the PHD2 peptidyl substrates is important for interactions with the ß2ß3/loop region of PHD2, and the C-terminal side of the substrates is important for interactions with the  $\alpha 4$  helix of PHD2.  $^{[73-75]}$  It was also shown that it is possible to tune the hydroxylation of PHD2

peptidyl substrates by using CODD/NODD hybrids.[44] Using this information, we designed two CODD/NODD hybrid peptides by swapping the N- and C-terminal sides of the prolyl residue of CODD and NODD. The two peptides with the sequences DALTLLAAPYIPMDDDFQL and DLDLEMLAPYIPTIISLDF are denoted Hybrids #01 and #02, respectively. In Hybrid #01, the two leucine residues in -5 and -2 positions relative to the proline in the conserved LxxLAP motif were replaced by threonine and alanine, respectively. Huang et al. showed that PHDs could tolerate substrates with substitution in these positions.<sup>[76]</sup> Indeed, using the MADLI-TOF MS assay, it was found that both Hybrids #01 and #02 were substrates of PHD2. The  $K_{\rm M}$  values of Hybrids #01 and #02 were found to be 36.6  $\pm$  5.3  $\mu$ M and 130.6  $\pm$  31.6  $\mu$ M respectively (Table 1 and Figure S11). We then measured the IC<sub>50</sub> values of CORM-2 to PHD2 with Hybrids #01 and #02 as substrates. We found that the IC<sub>50</sub> value of CORM-2 roughly followed the same trend (Table 1 and Figure S10), i.e. the lower the  $K_{\rm M}$  value of the substrate, the higher the IC<sub>50</sub> value. In comparison, the IC<sub>50</sub> value of CORM-2 in the absence of peptidyl substrate (i.e. uncoupled turnover) was found to be 38.9 ± 1.1 µM (Table 1 and Figure S10).

In the consensus mechanism of 20G-dependent oxygenases (Figure S12), the binding of the primary substrate generates an open coordination site on the Fe(II) centre at the active site by displacing a water ligand.<sup>[38,77]</sup> Our results may therefore reflect how different peptidyl substrates may modulate the access and/or binding of gaseous molecules (e.g. O<sub>2</sub> and CO) to the metal ion at the active site of PHD2. CO prefers to bind to Fe(II) with a linear geometry. It is possible that different substrates may distort the binding geometry of CO to different extents. The ability of peptidyl substrates to modulate the binding of gaseous molecules to 2OG-dependent oxygenases has been observed before.<sup>[78]</sup> For example, Taabazuing and co-workers reported that, by using density functional theory (DFT) calculations, the binding of peptidyl substrates may dictate and distort the binding geometry of nitric oxide (NO) at the metal centre of factor inhibiting HIF (FIH), another 2OG-dependent oxygenase that catalyses the hydroxylation of HIF.<sup>[78]</sup> Indeed, we found that the addition of excess peptidyl substrate (CODD or NODD) may rescue the activity of PHD2 against the inhibition by CO (Figure S13), further supporting the hypothesis that peptidyl substrates may affect the binding of gaseous ligands to PHD2. However, further modelling and/or structural-based experiments are needed to fully understand this effect (e.g. whether the prolyl residue of CODD/NODD directly clashes with the binding of CO in the linear aeometry).

Overall, our work shows for the first time that CO is an inhibitor of HIF PHDs. There is evidence showing the involvement of CO in cellular oxygen sensing and signalling, including the stabilisation of HIF in normoxia.<sup>[23,24]</sup> Although previous work proposed that CO may induce stabilisation of HIF through a PHD-independent pathway (by increasing the interactions between HIF $\alpha$  and heat shock protein 90),<sup>[79]</sup> our results show that CO is an inhibitor of PHD2 and suggest that the PHD-dependent pathway could, at least in part, play a role in the stabilisation of HIF in normoxia. The regulation of PHDs and FIH by endogenously produced gasotransmitters such as CO and NO is an underexplored yet important area. It is already known that PHDs and FIH could be inhibited by NO.<sup>[80–82]</sup> There is also evidence that CO and NO may work together for their

anuscr

# COMMUNICATION

physiological functions.<sup>[83]</sup> The cellular concentration of CO was estimated to be ~2 nM,<sup>[84]</sup> which is similar to the values reported for NO (~100 pM to ~5 nM).[85] The reported IC<sub>50</sub> value of NO to PHD2 was 10 µM,<sup>[86]</sup> which is in the same range as the IC<sub>50</sub> values obtained for CORM-2 in this study. Although these values are higher than the cellular concentrations of CO and NO, it has also been hypothesised that local CO concentrations, for example in the immediate vicinity where CO is being produced, may be significantly higher than the background concentration so that signalling can be triggered.<sup>[84]</sup> Therefore, our results may open up a new area of study, in particular addressing how CO and NO may interfere with the activity of PHDs (and other 2OG-dependent oxygenases) as well as their physiological relevance. Finally, there is increasing interest to exploiting the therapeutic potential of CO in medicine.<sup>[87,88]</sup> The observation that CO may inhibit PHDs, which themselves are molecular targets for the treatment of a number of diseases, may open up further avenues for the development of CO therapies.

**Table 1.** Table summarising the  $K_M$  and  $K_D$  values of the various peptidyl substrates of PHD2, and the IC<sub>50</sub> values of CORM-2 when these peptides were used as substrates.

Peptidyl substrate	<i>К</i> м / μМ	<i>K</i> <sub>D</sub> / μΜ	IC <sub>50</sub> / µM
1 9 000			
CODD: DLDLEMLAPYIPMDDDFQL	5.2 ± 1.3	~14 <sup>[a]</sup>	142.5 ± 1.8
NODD: DALTLLAPAAGDTIISLDF	107.1 ± 29.7	~85 <sup>[a]</sup>	34.2 ± 1.3
Hybrid #01: DALTLLAAPYIPMDDDFQL	36.6 ± 5.3	N/A	46.8 ± 1.1
Hybrid #02: DLDLEMLAPYIPTIISLDF	130.6 ± 31.6	N/A	24.4 ± 1.2
No substrate (uncoupled turnover)	N/A	N/A	38.9 ± 1.1

[a] K<sub>D</sub> values for CODD and NODD were obtained from reference [45].

#### Acknowledgements

We thank the University of Auckland for funding. N.M.M. was supported by grant-in-aid scholarships funded by the School of Chemical Sciences of the University of Auckland. N.N. was supported by a Science Education Scholarship funded by the Majlis Amanah Rakyat Agency (MARA). P.G.V. was supported by a doctoral scholarship funded by the MacDiarmid Institute for Advanced Materials and Nanotechnology, A.Z.I. would like to acknowledge the financial support from the Malaysian Ministry of Higher Education (MOHE) under the Academic Training Scheme SLAB-KPT/2013-2016. M. H. is supported by a Sir Charles Hercus Health Research Fellowship through the Health Research Council of New Zealand. We thank Dr Ryan Dixon (NMR spectroscopy), Martin Middleditch (protein mass spectrometry), Dr Gus Grey (MALDI-TOF mass spectrometry) and Dr Mansa Nair (non-denaturing nano-ESI mass spectrometry) for technical support. There are no conflicts to declare.

Keywords: gasotransmitter • PHD2 • CORM-2 • HIF $\alpha$  • carbon monoxide

- [1] R. F. Coburn, Prev. Med. 1979, 8, 310–322.
- [2] T. Sjöstrand, Acta Physiol. Scand. 1952, 26, 338.
- [3] T. Sjöstrand, Acta Physiol. Scand. 1950, 22, 137–141.
- [4] J. Muñoz-Sánchez, M. E. Chánez-Cárdenas, Oxid. Med. Cell. Longev. 2014, 2014, 604981.
- [5] L. Wu, R. Wang, *Pharmacol. Rev.* **2005**, 57, 585–630.
- [6] A. A. Untereiner, L. Wu, R. Wang in Gasotransmitters: Physiology and Pathophysiology (Eds.: A. Hermann, G. F. Sitdikova, T. M. Weiger), Springer, 2002, pp. 37–70.
- [7] B. E. Mann in Medicinal Organometallic Chemistry. Topics in Organometallic Chemistry, Vol. 32 (Eds.: G. Jaouen, N. Metzler-Nolte), Springer, 2010, pp. 247–285.
- [8] H. P. Kim, S. W. Ryter, A. M. K. Choi, Annu. Rev. Pharmacol. Toxicol. 2006, 46, 411–449.
- [9] R. Motterlini, R. Foresti, Am. J. Physiol. Cell Physiol. 2017, 312, C302– C313.
- [10] K. A. Hanafy, J. Oh, L. E. Otterbein, Curr. Pharm. Des. 2013, 19, 2771– 2775.
- [11] T. M. Dawson, S. H. Snyder, J. Neurosci. 1994, 14, 5147–5159.
- [12] R. Wang, Can. J. Physiol. Pharmacol. 1998, 76, 1–15.
- [13] R. Wang, Z. Wang, L. Wu, *Br. J. Pharmacol.* **1997**, *121*, 927–934.
- [14] S. W. Ryter, Arch. Biochem. Biophys. 2019, 678, 108186.
- [15] C. S. F. Queiroga, A. S. Almeida, H. L. A. Vieira, *Biochem. Res. Int.* 2012, 749845.
- [16] N. Schallner, L. E. Otterbein, Front. Physiol. 2015, 6, 17.
- [17] C. E. N.Reiter, A. I. Alayash, FEBS Open Bio 2012, 2, 113–118.
- [18] B. Y. Chin, G. Jiang, B. Wegiel, H. J. Wang, T. MacDonald, X. C. Zhang, D. Gallo, E. Cszimadia, F. H. Bach, P. J. Lee, L. E. Otterbein, *Proc. Natl. Acad. Sci. U.S.A.* 2007, 104, 5109–5114.
- [19] T. Morita, S. A. Mitsialis, H. Koike, Y. Liu, S. Kourembanas, J. Biol. Chem. 1997, 272, 32804–32809.
- [20] M. L. Dallas, J. P. Boyle, C. J. Milligan, R. Sayer, T. L. Kerrigan, C. McKinstry, P. Lu, J. Mankouri, M. Harris, J. L. Scragg, H. A. Pearson, C. Peers, *FASEB J.* 2011, *25*, 1519–1530.
- [21] C. A. Piantadosi, J. Zhang, E. D. Levin, R. J. Folz, D. E. Schmechel, *Exp. Neurol.* 1997, 147, 103–114.
- [22] A. S. Almeida, C. S. Queiroga, M. F. Sousa, P. M. Alves, H. L. Vieira, J. Biol. Chem. 2012, 287, 10761–10770.
- [23] N. R. Prabhakar, J. L. Dinerman, F. H. Agani, S. H. Snyder, Proc. Natl. Acad. Sci. U.S.A. 1995, 92, 1994–1997.
- [24] G. Yuan, C. Vasavda, Y. J. Peng, V. V. Makarenko, G. Raghuraman, J. Nanduri, M. M. Gadalla, G. L. Semenza, G. K. Kumar, S. H. Snyder, N. R. Prabhakar, Sci. Signal. 2015, 8, ra37.
- [25] Y. K. Choi, C.-K. Kim, H. Lee, D. Jeoung, K.-S. Ha, Y.-G. Kwon, K.-W. Kim, Y.-M. Kim, J. Biol. Chem. 2010, 285, 32116–32125.
- [26] L. E. Huang, W. G. Willmore, J. Gu, M. A. Goldberg, H. F. Bunn, J. Biol. Chem. 1999, 274, 9038–9044.
- [27] Y. Liu, H. Christou, T. Morita, E. Laughner, G. L. Semenza, S. Kourembanas, J. Biol. Chem. 1998, 273, 15257–15262.
- [28] A. C. R. Epstein, J. M. Gleadle, L. A. McNeill, K. S. Hewitson, J. O'Rourke, D. R. Mole, M. Mukherji, E. Metzen, M. I. Wilson, A. Dhanda, Y. M. Tian, N. Masson, D. L. Hamilton, P. Jaakkola, R. Barstead, J. Hodgkin, P. H. Maxwell, C. W. Pugh, C. J. Schofield, P. J. Ratcliffe, *Cell* 2001, *107*, 43– 54.
- [29] N. Masson, P. J. Ratcliffe, J. Cell Sci. 2003, 116, 3041–3049.
- [30] C. J. Schofield, P. J. Ratcliffe, *Nat. Rev. Mol. Cell Biol.* 2004, 5, 343–354.
- [31] W. G. Kaelin, Annu. Rev. Biochem. 2005, 74, 115–128.
- [32] G. H. Fong, K. Takeda, *Cell Death Differ.* **2008**, *15*, 635–641.
- [33] N. Yang, H. Su, T. Soga, K. R. Kranc, P. J. Pollard, *Hypoxia* 2014, 2, 127–142.
- [34] J. I. Bárdos, M. Ashcroft, Biochim. Biophys. Acta 2005, 1755, 107–120.
- [35] M. J. Ryle, R. P. Hausinger, Curr. Opin. *Chem. Biol.* **2002**, 6, 193–201.
- [36] R. P. Hausinger, *Crit. Rev. Biochem. Mol. Biol.* **2004**, 39, 21–68.

### COMMUNICATION

- [37] M. S. Islam, T. M. Leissing, R. Chowdhury, R. J. Hopkinson, C. J. Schofield, Annu. Rev. Biochem. 2018, 87, 585–620.
- [38] S. Martinez, R. P. Hausinger, J. Biol. Chem. 2015, 290, 20702–20711.
- [39] R. K. Bruick, S. L. McKnight, *Science* **2001**, 294, 1337–1340.
- [40] N. Masson, C. Willam, P. H. Maxwell, C. W. Pugh, P. J. Ratcliffe, *EMBO J.* 2001, 20, 5197–5206.
- [41] Y.-M. Tian, K. K. Yeoh, M. K. Lee, T. Eriksson, B. M. Kessler, H. B. Kramer, M. J. Edelmann, C. Willam, C. W. Pugh, C. J. Schofield, P. J. Ratcliffe, *J. Biol. Chem.* **2011**, *286*, 13041–13051.
- [42] E. Flashman, E. A. L. Bagg, R. Chowdhury, J. Mecinović, C. Loenarz, M. A. McDonough, K. S. Hewitson, C. J. Schofield, *J. Biol. Chem.* 2008, 283, 3808–3815.
- [43] S. Pektas, M. J. Knapp, J. Inorg. Biochem. 2013, 126, 5–60.
- M. I. Abboud, R. Chowdhury, I. K. H. Leung, K. Lippl, C. Loenarz, T. D.
   W. Claridge, C. J. Schofield, *ChemBioChem* 2018, 19, 2262–2267.
- [45] I. K. H. Leung, E. Flashman, K. K. Yeoh, C. J. Schofield, T. D. W. Claridge, *J. Med. Chem.* **2010**, 53, 867–875.
- [46] R. Motterlini, *Biochem. Soc. Trans.* 2007, 35, 1142–1146.
- [47] B. J. Lemon, J. W. Peters, *Biochemistry* **1999**, *40*, 12969–12973.
- [48] Z. Chen, B. J. Lemon, S. Huang, D. J. Swartz, J. W. Peters, K. A. Bagley, *Biochemistry* 2002, 41, 2036–2043.
- [49] S. T. Stripp, G. Goldet, C. Brandmayr, O. Sanganas, K. A. Vincent, M. Haumann, F. A. Armstrong, T. Happe, *Proc. Natl. Acad. Sci. U.S.A.* 2009, 106, 17331–17336.
- [50] H. Ogata, Y. Mizoguchi, N. Mizuno, K. Miki, S.-I. Adachi, N. Yasuoka, T. Yagi, O. Yamauchi, S. Hirota, Y. Higuchi, *J. Am. Chem. Soc.* 2002, *124*, 11628–11635.
- [51] A. L. DeLacey, C. Stadler, V. M. Fernandez, E. C. Hatchikian, H.-J. Fan, S. Li, M. B. Hall, *J. Biol. Inorg. Chem.* **2002**, *7*, 318–326.
- [52] H. A. Kuiper, K. Lerch, M. Brunori, A. Finazzi Agrò, FEBS Lett. 1980, 111, 232–234.
- [53] A. Finazzi-Agrò, L. Zolla, L. Flamigni, H. A. Kuiper, M. Brunori, *Biochemistry* 1982, 21, 415–418.
- [54] H. van der Deen, H. Hoving, Biophys. Chem. 1979, 9,169–179
- [55] R. J. Appelhoff, Y. M. Tian, R. R. Raval, H. Turley, A. L. Harris, C. W. Pugh, P. J. Ratcliffe, J. M. Gleadle, *J. Biol. Chem.* **2004**, *279*, 38458– 38465.
- [56] E. Berra, E. Benizri, A. Ginouvès, V. Volmat, D. Roux, J. Pouysségur, *EMBO J.* **2003**, *22*, 4082–4090.
- [57] E. Flashman, L. M. Hoffart, R. B. Hamed, J. M. Bollinger Jr, C. Krebs, C. J. Schofield, *FEBS J.* 2010, 277, 4089–4099.
- [58] C. C. Romão, W. A. Blättler, J. D. Seixas, G. J. L. Bernardes, *Chem. Soc. Rev.* 2012, *41*, 3571–3583.
- [59] R. Motterlini, J. E. Clark, R. Foresti, P. Sarathchandra, B. E. Mann, C. J. Green, *Circ. Res.* 2002, 90, e17–e24.
- [60] A. C. Kautz, P. C. Kunz, C. Janiak, *Dalton Trans.* 2016, *45*, 18045–18063.
   [61] A. J. Atkin, J. M. Lynam, B. E. Moulton, P. Sawle, R. Motterlini, N. M.
- Boyle, M. T. Pryce, I. J. S. Fairlamb, *Dalton Trans.* 2011, 40, 5755–5761.
  [62] R. Sekirnik, N. R. Rose, J. Mecinović, C. J. Schofield, *Metallomics* 2010,
- 2, 397–399.
- [63] J. Mecinović, R. Chowdhury, B. M. R. Liénard, E. Flashman, M. R. G. Buck, N. J. Oldham, C. J. Schofield, *ChemMedChem* 2008, 3, 569–572.
- [64] V. G. Nielsen, Int. J. Mol. Sci. 2020, 21, 2970.
- [65] K. Magierowska, M. Magierowski, M. Hubalewska-Mazgaj, J. Adamski, M. Surmiak, Z. Sliwowski, S. Kwiecien, T. Brzozowski, *PLoS One* 2015, 10, e0140493.
- [66] Y. Wei, P. Chen, M. de Bruyn, W. Zhang, E. Bremer, W. Helfrich, BMC Gastroenterol. 2010, 10, 42.
- [67] I. C. Winburn, K. Gunatunga, R. D. McKernan, R. J. Walker, I. A. Sammut, J. C. Harrison, *Basic Clin. Pharmacol. Toxicol.* **2012**, *111*, 31–41.
- [68] P. K. Witting, A. G. Mauk, J. Biol. Chem. 2001, 276, 16540–16547.
- [69] J. S. Blanchard, S. Englard, A. Kondo, Arch. Biochem. Biophys. 1982, 219, 327–334.
- [70] T. Santos-Silva, A. Mukhopadhyay, J. D. Seixas, G. J. L. Bernardes, C. C. Romão, M. J. Romão, *J. Am. Chem. Soc.* 2011, *133*, 1192–1195.

- [71] G. Gessner, N. Sahoo, S. M. Swain, G. Hirth, R. Schönherr, R. Mede, M. Westerhausen, H. H. Brewitz, P. Heimer, D. Imhof, T. Hoshi, S. H. Heinemann, *Eur. J. Pharmacol.* 2017, *815*, 33–41.
- [72] X. Lin, W. Zhao, X. Wang, J. Mass. Spectrom. 2020, 45, 618–626.
- [73] D. Villar, A. Vara-Vega, M. O. Landázuri, L. Del Peso, *Biochem. J.* 2007, 408, 231–240.
- [74] R. Chowdhury, M. A. McDonough, J. Mecinović, C. Loenarz, E. Flashman, K. S. Hewitson, C. Domene, C. J. Schofield, *Structure* 2009, 17, 981–989.
- [75] R. Chowdhury, I. K. H. Leung, Y.-M. Tian, M. I. Abboud, W. Ge, C. Domene, F.-X. Cantrelle, I. Landrieu, A. P. Hardy, C. W. Pugh, P. J. Ratcliffe, T. D. W. Claridge, C. J. Schofield, *Nat. Commun.* **2016**, *7*, 12673.
- [76] J. Huang, Q. Zhao, S. M. Mooney, F. S. Lee, J. Biol. Chem. 2002, 277, 39792–39800.
- [77] J. M. Bollinger Jr, W.-C. Chang, M. L. Matthews, R. J. Martinie, A. K. Boal, C. Krebs in 2-Oxoglutarate-Dependent Oxygenases (Eds.: R. P. Hausinger, C. J. Schofield), Royal Society of Chemistry, **2015**, pp. 95–122.
- [78] C. Y. Taabazuing, J. Fermann, S. Garman, M. J. Knapp, *Biochemistry* 2016, 55, 277–286.
- [79] Y. K. Choi, C.-K. Kim, H. Lee, D. Jeoung, K.-S. Ha, Y.-G. Kwon, K.-W. Kim, Y.-M. Kim, *J. Biol. Chem.* **2010**, *285*, 32116–32125.
- [80] U. Berchner-Pfannschmidt, H. Yamac, B. Trinidad, J. Fandrey, J. Biol. Chem. 2007, 282, 1788–1796.
- [81] L. E. Huang, W. G. Willmore, J. Gu, M. A. Goldberg, H. F. Bunn, J. Biol. Chem. 1999, 274, 9038–9044.
- [82] E. Metzen, J. Zhou, W. Jelkmann, J. Fandrey, B. Brüne, *Mol. Cell Biol.* 2003, 14, 3470–3481.
- [83] C. L. Hartsfield, Antioxid. Redox Signal. 2002, 4, 301–307.
- [84] D. G. Levitt, M. D. Levitt, Clin. Pharmacol. 2015, 7, 37–56.
- [85] C. N. Hall, J. Garthwaite, *Nitric Oxide* **2009**, *21*, 92–103.
- [86] R. Chowdhury, E. Flashman, J. Mecinović, H. B. Kramer, B. M. Kessler, Y. M. Frapart, J.-L. Boucher, I. J. Clifton, M. A. McDonough, C. J. Schofield, J. Mol. Biol. 2011, 410, 268–279.
- [87] R. Motterlini, L. E. Otterbein, Nat. Rev. Drug Discov. 2010, 9, 728–743.
- [88] S. W Ryter, L. E. Otterbein, *Bioessays* 2004, 26, 270–280.

# COMMUNICATION

#### Entry for the Table of Contents (Please choose one layout)

Layout 1:

# COMMUNICATION

Carbon monoxide (CO) is an endogenously produced gasotransmitter that is involved in a range of physiological roles including the stabilisation of HIF, a transcription factor that regulates oxygen homeostasis. Herein, by using CO- releasing molecule-2 as a CO donor, we showed that CO is an inhibitor of HIF PHD2, one of the most important oxygen sensing enzymes in humans that regulates HIF stability at the protein level.	Co-releasing (CORN-2) $e^{-C}$	Author(s), Corresponding Author(s)* Page No. – Page No. Title
Layout 2:		
COMMUNICATION		
		Author(s), Corresponding Author(s)*
((Insert TOC Gra	phic here))	Page No. – Page No. Title
		The
Text for Table of Contents		