

RESEARCHSPACE@AUCKLAND

http://researchspace.auckland.ac.nz

ResearchSpace@Auckland

Copyright Statement

The digital copy of this thesis is protected by the Copyright Act 1994 (New Zealand).

This thesis may be consulted by you, provided you comply with the provisions of the Act and the following conditions of use:

- Any use you make of these documents or images must be for research or private study purposes only, and you may not make them available to any other person.
- Authors control the copyright of their thesis. You will recognise the author's right to be identified as the author of this thesis, and due acknowledgement will be made to the author where appropriate.
- You will obtain the author's permission before publishing any material from their thesis.

To request permissions please use the Feedback form on our webpage. http://researchspace.auckland.ac.nz/feedback

General copyright and disclaimer

In addition to the above conditions, authors give their consent for the digital copy of their work to be used subject to the conditions specified on the <u>Library Thesis Consent Form</u> and <u>Deposit Licence</u>.

SURFACE PROPERTIES AND FINISHING OF ALUMINIUM CASTING ALLOYS

CHUONG LUU NGUYEN

A thesis submitted in fulfilment of the requirements for the degree of Doctor of Philosophy in Chemistry (Materials Science)

The University of Auckland, 2009

ABSTRACT

Aluminium alloys are materials of huge practical importance. However their use is dependent on surface oxides and hydroxides which are critical in protecting the highly reactive underlying metal. The stability and integrity of the oxides and hydroxides are also crucial in finishing and bonding applications. Better understanding of these surfaces has significant implications in enhancing their application.

LM6 and LM25 aluminium-silicon casting alloys were studied as these materials show a particularly inhomogeneous phase structure and complex surface behaviour. This complexity is of fundamental interest and leads to considerable practical difficulties, especially in surface finishing. The surfaces were characterised, subjected to thermal treatments and modified with Ion Assisted Deposition coatings of TiN. A characterisation method for these surfaces was also developed based around the layered structure of aluminium hydroxides.

It was observed that the thermally induced surface segregation of minor elements, such as Mg and Na, is availability-limited. Surface concentrations of these elements are determined by the net effect of enriching via surface segregation and depleting through surface evaporation. The inhomogeneous phase structure of the alloys used in this study enables the observation of two migration processes driven by different forces. Below the oxide dominated surface layer, the migration of Mg is driven by chemical potential gradient and is primarily perpendicular to the surface. Closer to the surface, concentration driven horizontal diffusion of the element occurs.

The deposition of a thin TiN layer has been used to probe the interface. While the deposition conditions of TiN coatings affect the chemistry of the coatings, substrate surface conditions determine how well bonding is achieved between the film and substrate. The height difference between silicon particles in the eutectic phase and the primary aluminium phase of the casting alloys provides surface roughening and optimal adhesion through mechanical interlocking with the coating. This height difference is achieved by preferentially removing the surface exposed primary aluminium phases with ion bombardment.

The nature of surface aluminium oxides and hydroxides and their modification, has been further probed by studying how repeating fragments in TOF-SIMS spectra of these compounds originate. Gibbsite powder and a bayerite layer on a metal surface have been examined, before and after intercalation into their layered structure. This suggests that the weak interlayer bonding can be manipulated and allows cleavage of the outermost octahedral layers under bombardment by energetic particles. This phenomenon provides a new and particularly sensitive characterisation method, using the relative intensities of the repeating fragments in the TOF-SIMS spectra.

To my parents

To my wife, a friend and a colleague of more than fifteen years

ACKNOWLEDGEMENTS

First and foremost, my gratitude goes to my supervisors – Professor James Metson and Associate Professor Margaret Hyland. They are always very supportive both inside and outside of the academic environment. While their knowledge and enthusiasm in scientific research are inspiring, their understandings of and sympathy to everyday situations are equally encouraging. I truly hope to continue receiving such supports in future.

It is an absolute privilege that the study was one of the projects in Light Metals Research Centre (LMRC), the University of Auckland. The most obvious benefits include the vast pool of knowledge at the centre, the accesses to various resources and the relationships built up over the years. I would like to express special thanks to Professor Mark Taylor, David Cotton, Marcus Gustafsson, Dr Ronny Etzion, Tania Groutso, Linus Perander, Dr Wei Zhang and Jenny Roper for their extra helps.

Extensive lab works have been carried out in both Department of Chemistry and Department of Chemical and Materials Engineering. For those, I would like to thank not only LMRC staff but also Anh Tran, Dr Geoffrey Waterhouse, Dr Michael Hodgson, Dr Bryony James, Dr Colin Doyle and Catherine Hobbis. Outside the university, experiments were done with helps from Dr Kathryn Prince and Armand Atanacio at Australian Nuclear Science and Technology Organisation (ANSTO), Associate Professor Paul Pigram and Dr Robert Jones at La Trobe University, Dr Ben Ruck, Andrew Preston and Bart Ludbrook at Victoria University of Wellington, Dr Kia Wallwork at the Australian Synchrotron, and Professor Paul Munroe and Dr Charlie Kong at the University of New South Wales.

Last but not least, financial assistances are greatly appreciated: the University of Auckland Doctoral Scholarship, funding from Auckland UniServices Limited and the Foundation for Research Science and Technology (FRST) in the Transforming Light Metals Project – Objective 2, Australian Institute of Nuclear Science and Engineering (AINSE) Grant AINGRA06127P, and the allocation from the Australian Synchrotron and New Zealand Synchrotron Group Limited for Proposal 927.

TABLE OF CONTENTS

ABSTRAC	Γ	ii
ACKNOW	LEDGEMENTS	v
TABLE OF	CONTENTS	vi
LIST OF FI	GURES	ix
LIST OF T	ABLES	xiii
Chapter 1.	INTRODUCTION	1
1.1. Ro	oles of surfaces and interfaces in materials performance	1
1.2. Ob	ojectives of this study	2
1.3. Re	search focus	3
Chapter 2.	LITERATURE REVIEW	5
2.1. Ch	aracterisation of aluminium surface	5
2.1.1.	Characterisation of surface acidity and basicity	5
2.1.2.	Characterisation of surface composition	10
2.2. Su	rface segregation in aluminium alloys	14
2.2.1.	Dynamics of surface segregation	15
2.2.2.	Kinetics of surface segregation and evaporation	19
2.3. Su	rface properties of aluminium alloys	23
2.3.1.	Surface properties due to composition	23
2.3.2.	Surface properties due to structure for aluminium oxides	24
2.4. Su	rface protection and finishing of aluminium alloys	25
2.4.1.	Hard coatings and titanium nitride	27
2.4.2.	Adhesion between coatings and substrates	29
Chapter 3.	EXPERIMENTAL METHODS	32
3.1. Ma	aterials	32
3.1.1.	Metal samples	32
3.1.2.	Powder samples	34
3.2. Sa	mple preparation	35
3.2.1.	Grinding and polishing	35
3.2.2.	Controlled atmosphere furnace	36
3.2.3.	Ion Assisted Deposition (IAD) of thin films	36
3.2.4.	Intercalation of aluminium hydroxide	37

3.3.	Ana	alysis techniques	38
3.3	.1.	X-ray Photoelectron Spectroscopy (XPS)	38
3.3 ED		Scanning Electron Microscopy/ Energy Dispersive Spectroscopy (SI 42	Ξ M /
3.3	.3.	Secondary Ion Mass Spectrometry (SIMS)	43
3.3	.4.	X-Ray Diffractometry (XRD)	45
3.3	.5.	Nano-Indentation	46
Chapter	4.	SURFACE SEGREGATION AND EVAPORATION OF TRACE	
ELEME	NTS	S IN ALUMINIUM CASTING ALLOYS	47
4.1.	Sur	face segregation and evaporation of Mg and Na in alloy LM6	47
4.2.	Sur	face segregation in alloy LM25	55
4.3.	Ma	thematical model for combined surface segregation and evaporation	57
4.3	.1.	Surface segregation equation	57
4.3	.2.	Surface evaporation equation	60
4.3	.3.	Combined mathematical model and fitting with experimental data	61
4.4.	Sur	nmary and discussion	65
Chapter	5.	SURFACE PROPERTIES OF LM6 AND LM25 ALUMINIUM	
ALLOY	S U	NDER THERMAL TREATMENT	67
5.1.	Ele	mental distribution in LM6 resulting from thermal treatment	68
5.2.	Ele	mental distribution in LM25 resulting from thermal treatment	74
5.3.	Oxi	de evolution on the surface of alloy LM6	79
5.4.	Sur	nmary and discussion	80
Chapter	6.	SURFACE FINISHING OF ALUMINIUM CASTING ALLOYS	82
6.1.	Cha	aracteristics of TiN films grown by IAD	82
6.2.	The	e interface between TiN films and aluminium substrates	87
6.3.	The	adhesion between TiN films and aluminium substrates	92
6.4.	Sur	nmary and discussion	94
Chapter	7.	SURFACE PROPERTIES AND COATING APPROACHES ON	
ALUMI	NIU	M AND ALUMINIUM BASED CASTING ALLOYS	95
7.1.	Che	emical, morphological and structural changes of modified gibbsites	96
7.1	.1.	Chemical changes	96
7.1	.2.	Morphological changes	98
7.1	.3.	Structural changes	.100
7.2.	The	e repeating fragments in TOF-SIMS spectra	.103

7.3.	Surface modification by ion bombardment	109
7.4.	Characterisation of surface species using repeating fragments	119
7.5.	Summary and discussion	124
Chapter	8. CONCLUSIONS	126
APPEN	DIX A – Plasmon peaks and the quantification of XPS data for aluminium	_
silicon a	ılloys	129
APPEN	DIX B – Solution to the equation for the combined effects of surface	
segregat	tion and evaporation	140
REFERI	ENCES	144

LIST OF FIGURES

Figure 2.1. Schematic of surface hydroxylation and charging in aqueous environment
Figure 2.2. Sites with lower acidity and higher basicity due to surrounding ions
Figure 2.3. Relative abundance (against 100 as the intensity of AlO ₂ -) of the most
intense TOF-SIMS negative fragments [26]
Figure 2.4. Diffusion model of Mg in Al alloys [42]17
Figure 2.5. The depth profile of segregant proposed by previous workers20
Figure 2.6. Universal segregation – evaporation curve of Al-08.%Mg alloy [2]22
Figure 2.7. Schematic of Ion Assisted Deposition (IAD)
Figure 2.8. Microscopic evaluation of scratch line (above) [91], and matching with the
plot of increasing load [94]
Figure 3.1. Inhomogeneous phase structures of LM6 (above) and LM25 casting alloys
observed in etched surfaces under SEM
Figure 4.1. SEM/ EDS line scan across an inter-metallic phase of LM6 before heating
48
Figure 4.2. SEM/ EDS spot scan on inter-metallic phase of LM6 after heating49
Figure 4.3. Typical DSIMS depth profiles of LM6 after heating
Figure 4.4. Typical XPS depth profile of LM6 after heating
Figure 4.5. Variation in the contents of minor elements in LM6 after heating52
Figure 4.6. Ternary phase diagram Al-Si-Mg [101]
Figure 4.7. TOF-SIMS elemental map of LM6 after heating
Figure 4.8. XPS narrow scan (a) Mg 2p; (b) C 1s55
Figure 4.9. Typical DSIMS profiles of LM25 before and after heating to 440 $^{\circ}$ C56
Figure 4.10. Diffusion model based on diffusion potential and a growing oxide layer
Figure 4.11. Fitted model for the film growth constant n at 400°C
Figure 4.12. Fitted model for Mg segregation in the temperature range between 320
and 480 °C
Figure 4.13. Fitted model for Na segregation in the temperature range between 320
and 480 °C

Figure 5.1. DSIMS depth profile of unheated LM6	69
Figure 5.2. DSIMS elemental maps of unheated LM6 showing images at three dept	ths
into the sample.	69
Figure 5.3. Oxide thicknesses calculated from Strohmeier's formula for LM6 samp	les
heated to 400°C for 60 minutes and 440°C for 360 minutes	70
Figure 5.4. DSIMS depth profile of LM6 heated to 400°C for 60 minutes	71
Figure 5.5. DSIMS elemental maps of LM6 heated to 400°C for 60 minutes	71
Figure 5.6. DSIMS depth profile of LM6 heated to 440°C for 360 minutes	72
Figure 5.7. DSIMS elemental maps of LM6 heated to 440°C for 360 minutes	73
Figure 5.8. DSIMS depth profile of LM25 unheated	75
Figure 5.9. DSIMS elemental maps of LM25 unheated	75
Figure 5.10. DSIMS depth profile of LM25 heated to 440°C for 60 minutes	76
Figure 5.11. DSIMS elemental maps of LM25 heated to 440°C for 60 minutes	77
Figure 5.12. DSIMS depth profile of LM25 heated to 440°C for 24 hours	78
Figure 5.13. DSIMS elemental maps of LM25 heated to 440°C for 24 hours	78
Figure 5.14. TOF-SIMS images of LM6 surface – sputter cleaned and re-oxidised i	in
high vacuum	79
Figure 6.1. TEM images of the cross sections of the coatings resulted from Runs #2	2
and #3	84
Figure 6.2. TEM/ EDS elemental mapping of the coating resulted from Run #2	84
Figure 6.3. XPS narrow scans of the Ti 2p and N 1s regions of IAD grown TiN file	ms.
	85
Figure 6.4. DSIMS depth profiles of TiN coatings on LM6 and LM25.	89
Figure 6.5. Schematic of the interface between a TiN coating and a substrate of an	
aluminium casting alloy	90
Figure 6.6. SEM images showing Si particles exposed on the substrate surfaces	91
Figure 6.7. Topography of the TiN coatings on different substrates	92
Figure 6.8. Scratch lines on the coatings (Run #3) on high purity aluminium (above	e),
and on LM25	93
Figure 7.1. XPS survey spectra of Fe-modified gibbsite (lower) and the original	97
Figure 7.2. XPS O 1s and Fe 2p narrow scans of Fe-modified gibbsite	98
Figure 7.3. SEM images of (a) original; (b) Na-modified; (c) Fe modified gibbsites	.99
Figure 7.4. XRD spectra of original and Fe-modified gibbsites	101
Figure 7.5. Rietveld refinement results of synchrotron diffraction data	102

Figure 7.6. TOF-SIMS spectrum of original gibbsite with clear intermediate peaks 105
Figure 7.7. TOF-SIMS spectrum of Na-modified gibbsite showing weaker
intermediate peaks
Figure 7.8. TOF-SIMS spectrum of Fe-modified gibbsite showing insignificant
intermediate peaks
Figure 7.9. Schematic of proposed origins of repeating fragments in TOF-SIMS
spectra of gibbsite
Figure 7.10. The effect of sputtering by Cs ⁺ on repeating patterns of hydroxylated and
modified surface of high purity aluminium
Figure 7.11. TOF-SIMS spectra of hydroxylated surface of high purity aluminium
before and after sputtering
Figure 7.12. TOF-SIMS spectra of hydroxylated and modified surface of high purity
aluminium before and after sputtering
Figure 7.13. TOF-SIMS spectrum of hydroxylated and modified surface of high
purity aluminium showing new families of repeating fragments114
Figure 7.14. TOF-SIMS spectra of hydroxylated surface of LM6 casting alloy before
and after sputtering
Figure 7.15. TOF-SIMS spectra of hydroxylated and modified surface of LM6 casting
alloy before and after sputtering
Figure 7.16. TOF-SIMS spectrum of hydroxylated and modified surface of LM6
casting alloy showing new families of repeating fragments
Figure 7.17. Exponentially fitted trends of abundance of $(Al_2O_3)_xAlO_2^-$ fragments for
gibbsite powders
Figure 7.18. Exponentially fitted trends of abundance of $(Al_2O_3)_xAlO_2^-$ fragments for
gibbsite powder (unmodified) and the hydroxylated and modified surfaces of high
purity aluminium and LM6 alloy (before sputtering)121
Figure 7.19. Exponentially fitted trends of abundance of (Al ₂ O ₃) _x AlO ₂ ⁻ fragments for
hydroxylated surface of high purity aluminium – before and after sputtering122
Figure 7.20. Exponentially fitted trends of abundance of (Al ₂ O ₃) _x AlO ₂ fragments for
hydroxylated and modified surface of high purity aluminium - before and after
sputtering
Figure 7.21. Exponentially fitted trends of abundance of $(Al_2O_3)_xAlO_2^-$ fragments for
hydroxylated surface of LM6 alloy – before and after sputtering123

Figure 7.22. Exponentially fitted trends of abundance of $(Al_2O_3)_xAlO_2^-$ fragments for hydroxylated and modified surface of LM6 alloy – before and after sputtering 124

LIST OF TABLES

Table 1.1. Designations of the alloys used in this study	2
Table 2.1. Ranges of binding energy reported for aluminium oxides and hydrox	ides 10
Table 2.2. Fragments inferred to appear in the TOF-SIMS spectra of oxide M-C	[24]
	12
Table 2.3. Summary of works done in surface segregation of Mg in Al alloys	18
Table 2.4. Acid – base properties of some oxides [66]	24
Table 2.5. Guide to selection of finishes for casting alloys [72]	26
Table 3.1. Bulk composition of LM6 and LM25 casting alloys (weight %)	32
Table 3.2. Bulk composition of LM6 and LM25 casting alloys (atomic %	
approximately)	32
Table 3.3. Parameters of TiN deposition runs	37
Table 3.4. Typical XPS analysis sequence	41
Table 4.1. Parameters of the mathematical model for Mg segregation	64
Table 4.2. Parameters of the mathematical model for Na segregation and evapor	ration
	65
Table 6.1. Identification of the deconvoluted peaks in Figure 6.3.	86
Table 6.2. Calculation of Ti:N stoichiometry for TiN (excluding oxynitride)	86
Table 6.3. Calculation of Ti:N stoichiometry for TiOxNy (excluding nitride)	87
Table 6.4. Characteristics of the TiN coatings grown by IAD	87
Table 7.1. Summary of the Rietveld refinement results of gibbsites	103